

Surface Micromachining

Chang Liu

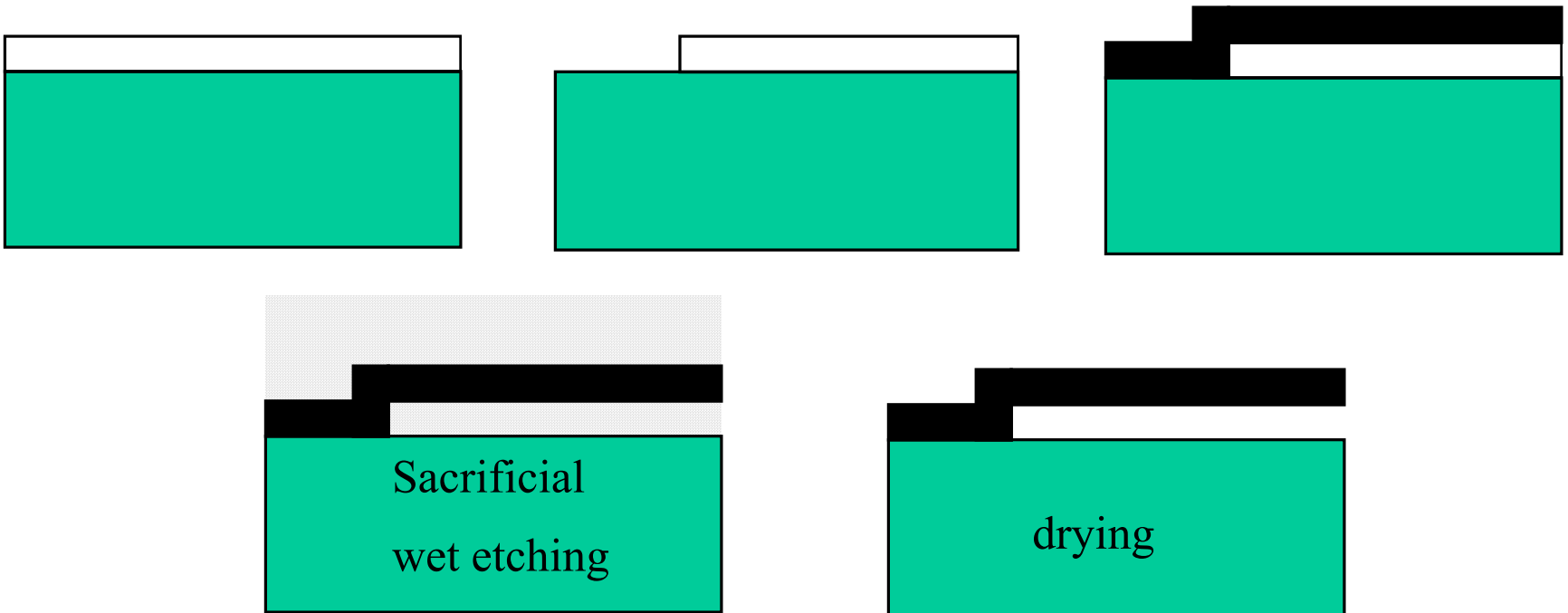
Micro Actuators, Sensors, Systems Group
University of Illinois at Urbana-Champaign

Outline

- Definition of surface micromachining
- Most common surface micromachining materials - polysilicon and silicon oxide
 - LPCVD deposition of polysilicon, silicon nitride, and oxide
 - plasma etching for patterning structural layer
 - micromachined hinges - fabrication process and assembly technique
 - micromachined dimples and scratch drive actuators
- Other sacrificial processing systems
 - metal sacrificial layer, plastics materials, etc.
- Stiction and anti-stiction solutions
- Multi User MEMS Process (MUMPS)
 - process definition and layer naming conventions

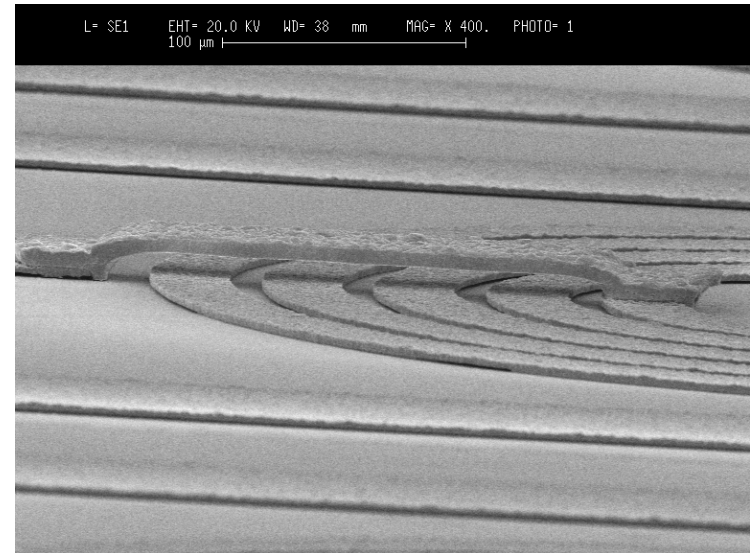
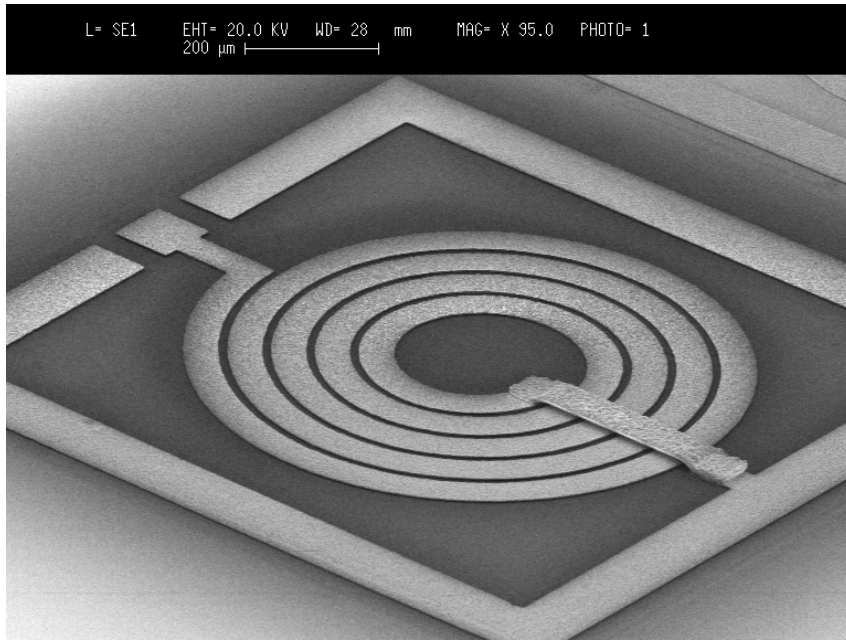
Basic Sacrificial Layer Processing

- Step 1: Deposition of sacrificial layer
- Step 2: patterning of the sacrificial layer
- Step 3: deposit structural layer (conformal deposition)
- Step 4: liquid phase removal of sacrificial layer
- Step 5: removal of liquid - drying.



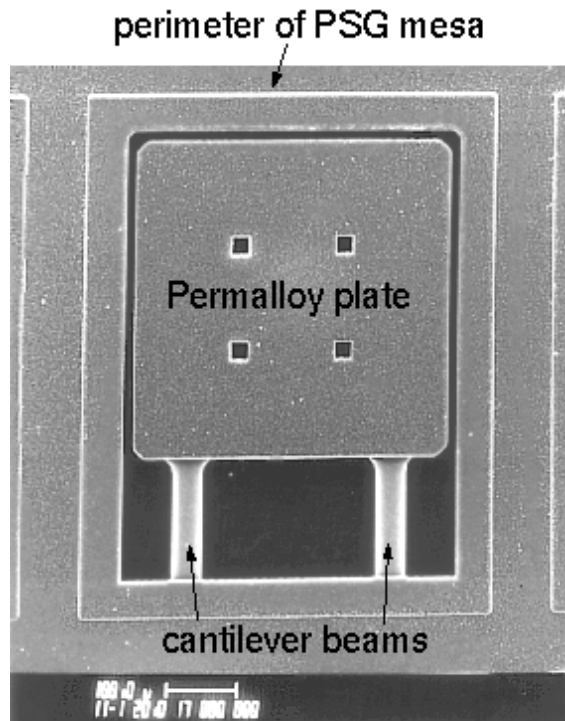
Surface Micromachined Inductor

- Air bridge can be formed using sacrificial etching.

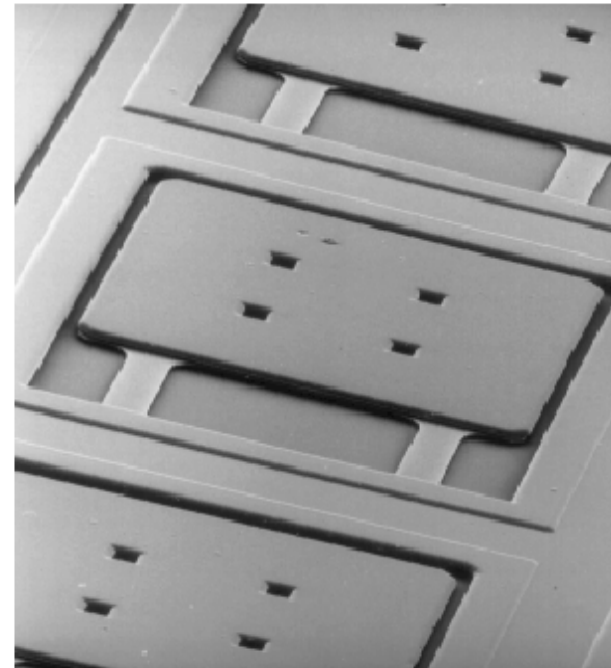


Typical Device Realized by Surface Micromachining

- Etch holes are required to reduce the time for removing sacrificial layer underneath large-area structures.

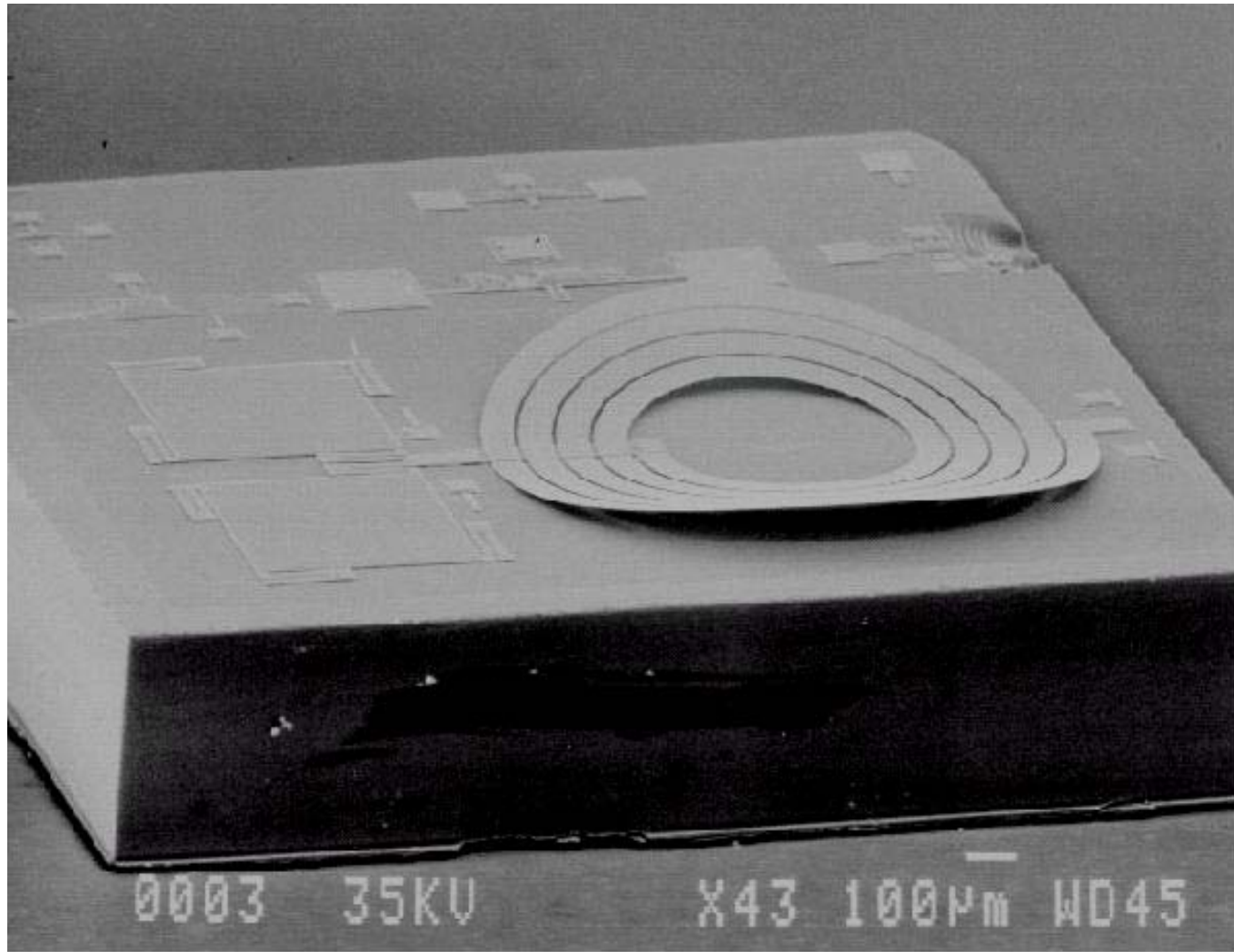


(a)

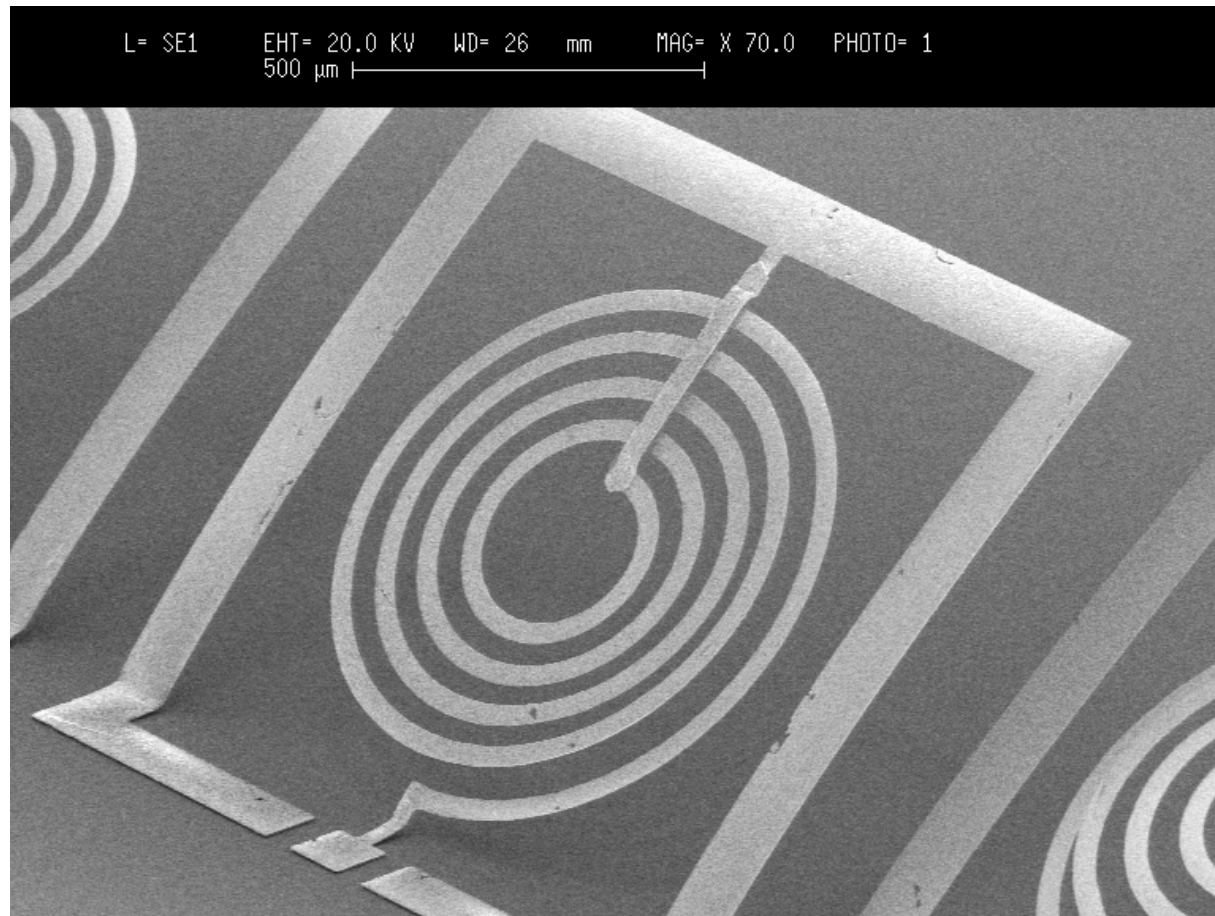


(b)

Inductor - By Lucent Technologies

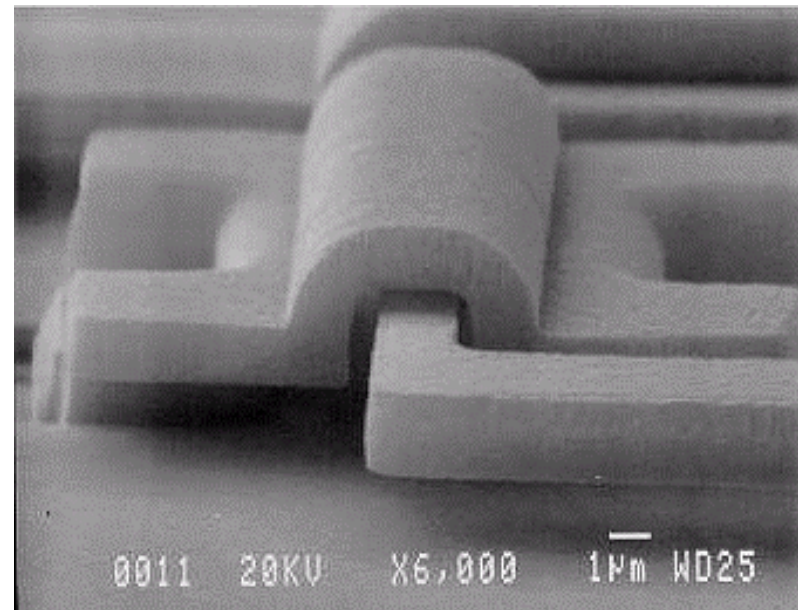
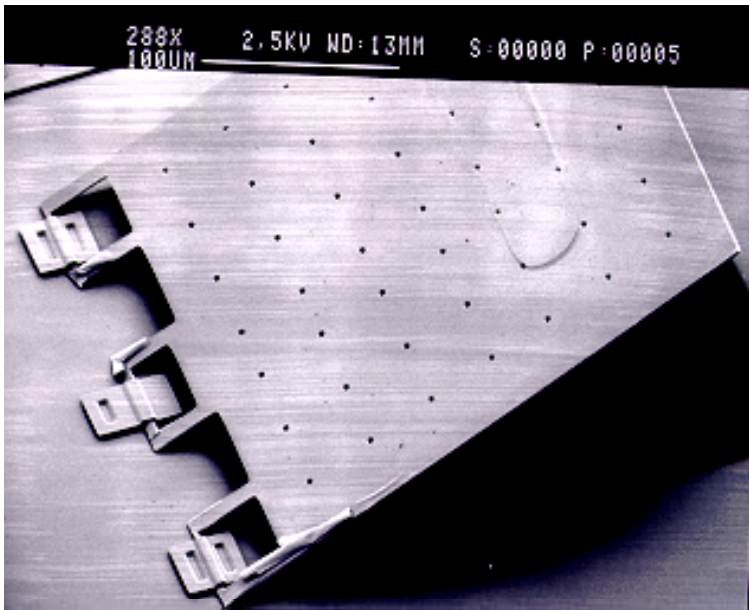


Surface Micromachined, Out of Plane Structures



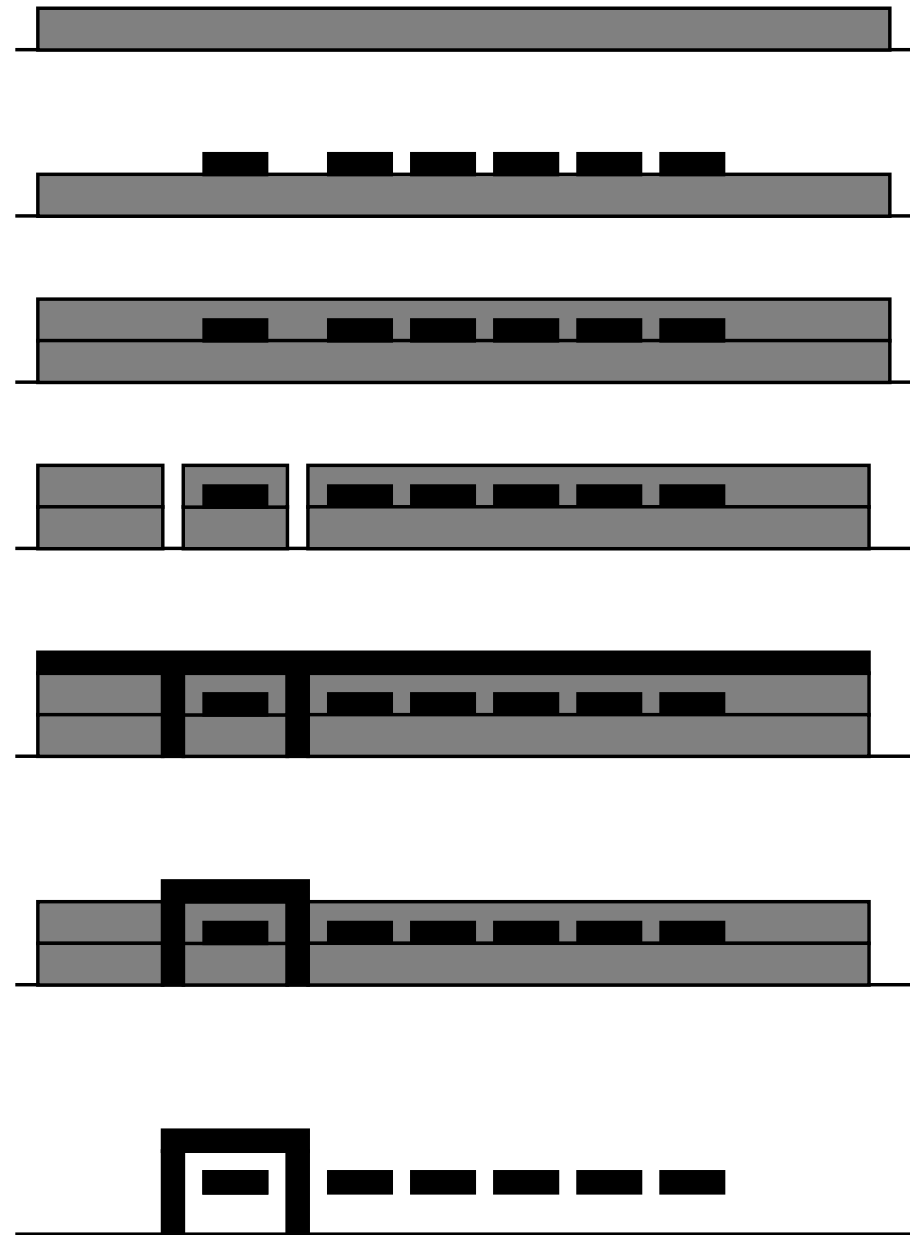
Hinges

- Used in micro optics component assembly.



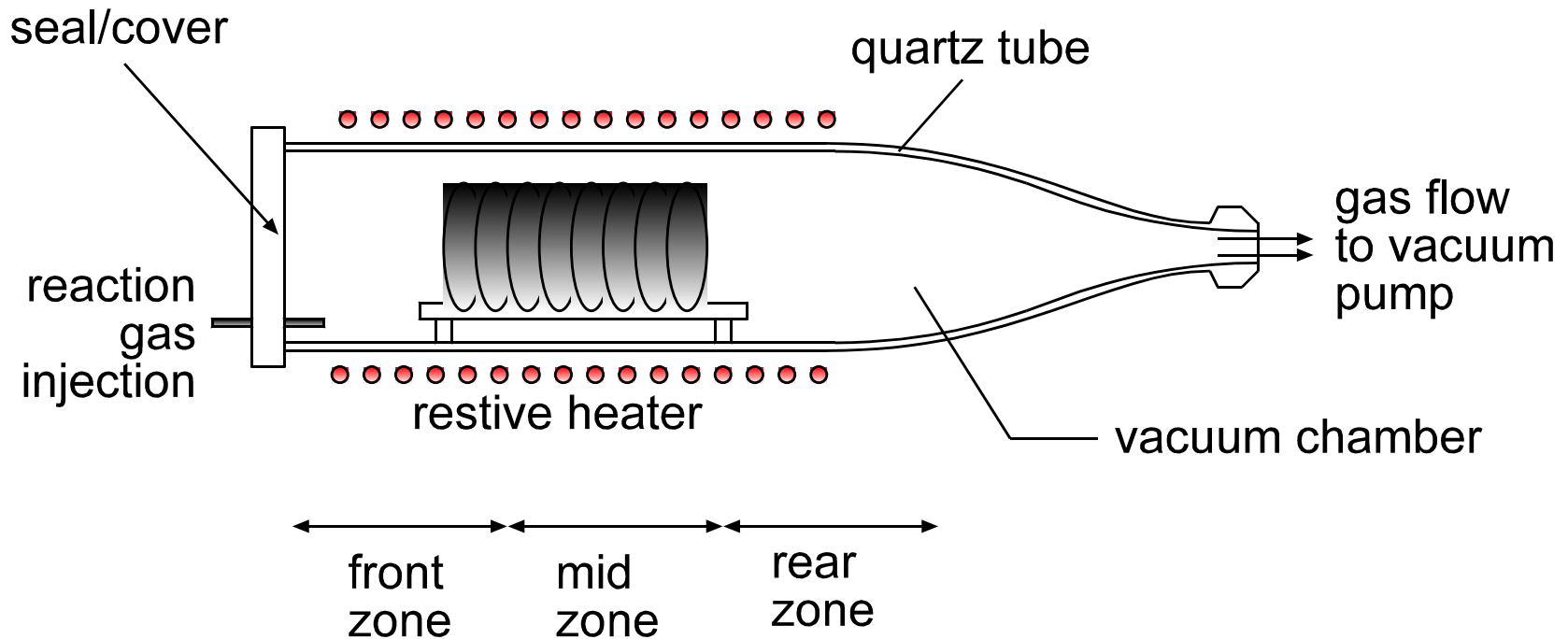
Hinge Fabrication

- Step 1: deposition of sacrificial layer.
- Step 2: deposition of structural layer.
- Step 3: deposition of second sacrificial layer.
- Step 4: etching anchor to the substrate.
- Step 5: deposition of second structural layer.
- Step 6: patterning of second structural layer
- Step 7: Etch away all sacrificial layer to release the first structural layer.

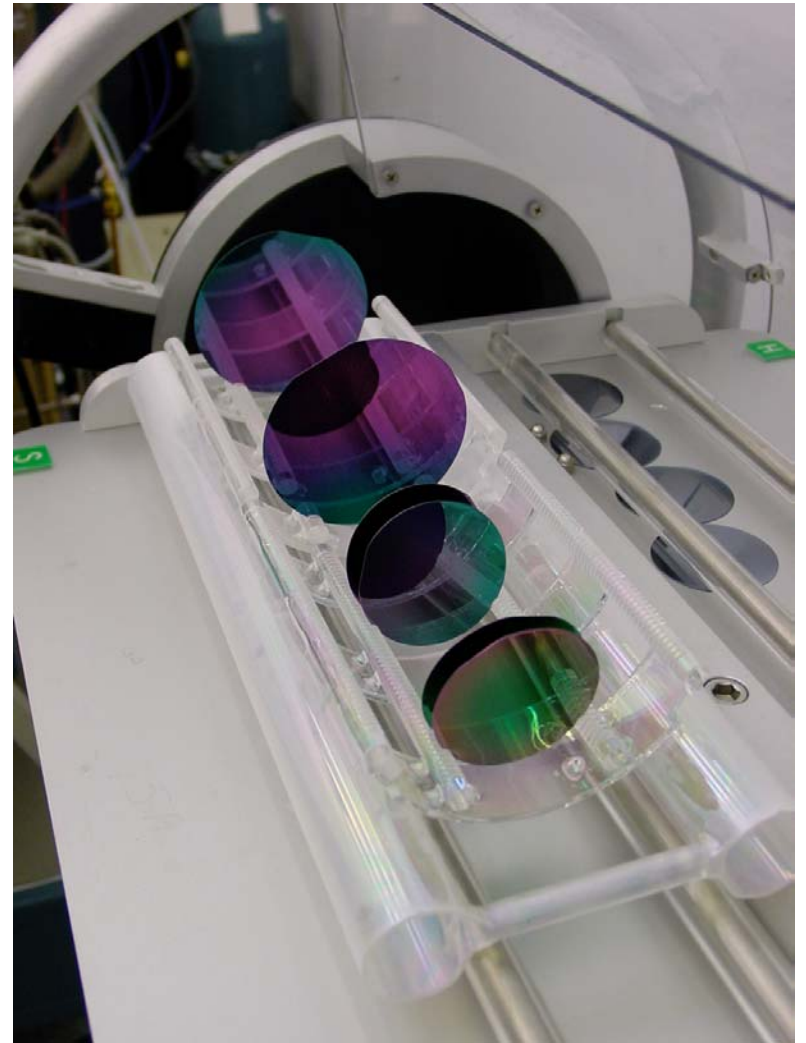


LPCVD Process

- Temperature range 500-800 degrees
- Pressure range 200 - 400 mtorr (1 torr = 1/760 ATM)
- Gas mixture: typically 2-3 gas mixture
- Particle free environment to prevent defects on surface (pin holes)



A Laboratory LPCVD Machine



LPCVD Recipes for Silicon Nitride, Polysilicon, and Oxide

- **Polycrystalline silicon**

- Polysilicon is deposited at around 580-620 °C and can withstand more than 1000 °C temperature. The deposition is conducted by decomposing silane (SiH₄) under high temperature and vacuum ($\text{SiH}_4 \rightarrow \text{Si} + 2\text{H}_2$).
- Polysilicon is used extensively in IC - transistor gate

- **Silicon nitride**

- Silicon nitride is nonconducting and has tensile intrinsic stress on top of silicon substrates. It is deposited at around 800 °C by reacting silane (SiH₄) or dichlorosilane (SiCl₂H₂) with ammonia (NH₃) - $\text{SiH}_4 + \text{NH}_3 \rightarrow \text{Si}_x\text{N}_y + \text{H}_2$.

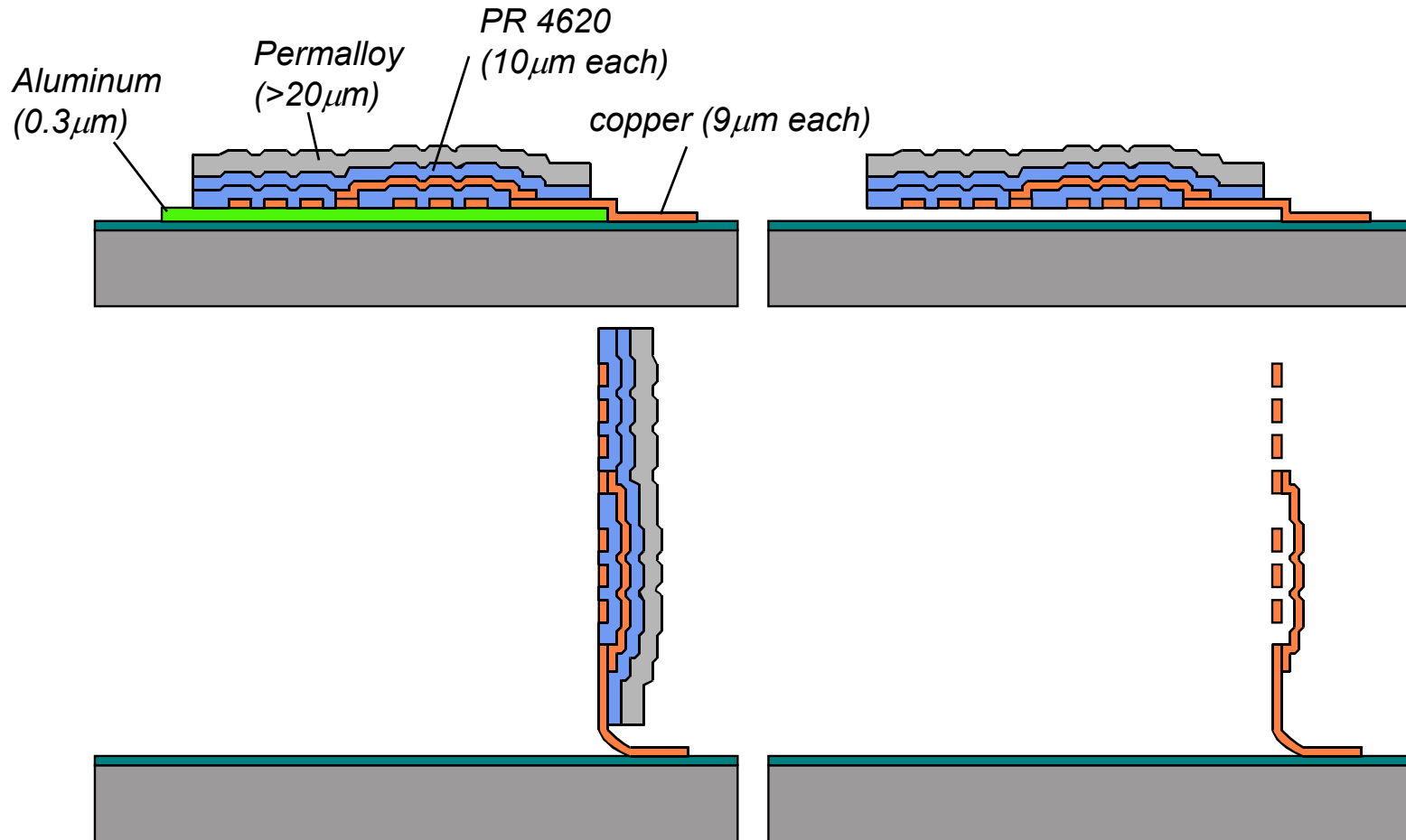
- **Silicon oxide**

- The PSG is known to reflow under high temperature (e.g. above 900 °C); it is deposited under relatively low temperature, e.g. 500 °C by reacting silane with oxygen ($\text{SiH}_4 + \text{O}_2 \rightarrow \text{SiO}_2 + 2\text{H}_2$). PSG can be deposited on top of Al metallization.
- Silicon oxide is used for sealing IC circuits after processing.
- The etch rate of HF on oxide is a function of doping concentration.

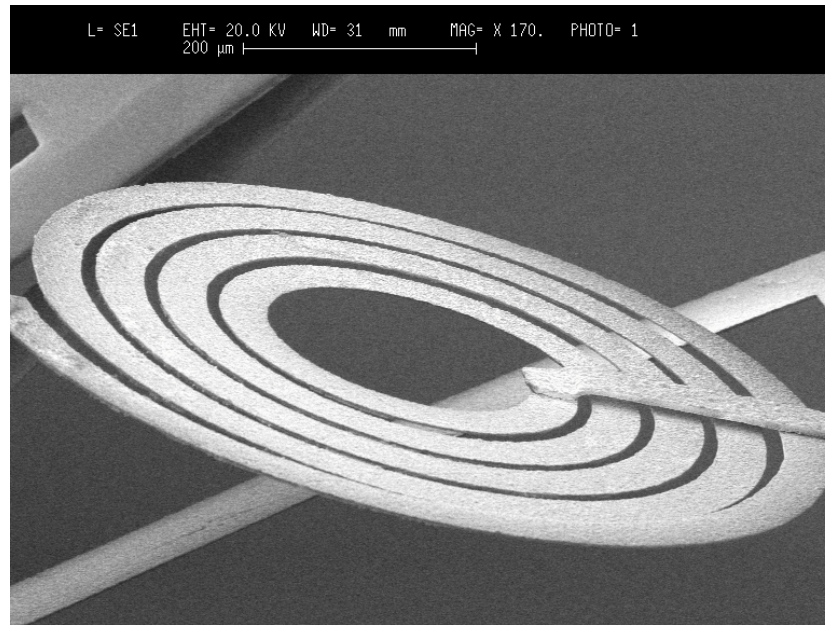
Other Structural or Sacrificial Materials

- **Structural layers**
 - evaporated and sputtered metals such as Gold, Copper
 - electroplated metal (such as NiFe)
 - plastic material (CVD plastic)
 - silicon (such as epitaxy silicon or top silicon in SOI wafer)
- **Sacrificial layers**
 - photoresist, polyimide, and other organic materials
 - copper
 - copper can be electroplated or evaporated, and is relatively inexpensive.
 - Oxide by plasma enhanced chemical vapor deposition (PECVD)
 - PECVD is done at lower temperature, with lower quality. It is generally undoped.
 - Thermally grown oxide
 - relatively low etch rate in HF.
 - Silicon or polysilicon
 - removed by gas phase silicon etching

Metal Sacrificial Layers

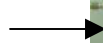


Out Of Plane Devices



A PECVD Machine

Processing
gases



Reaction
chamber

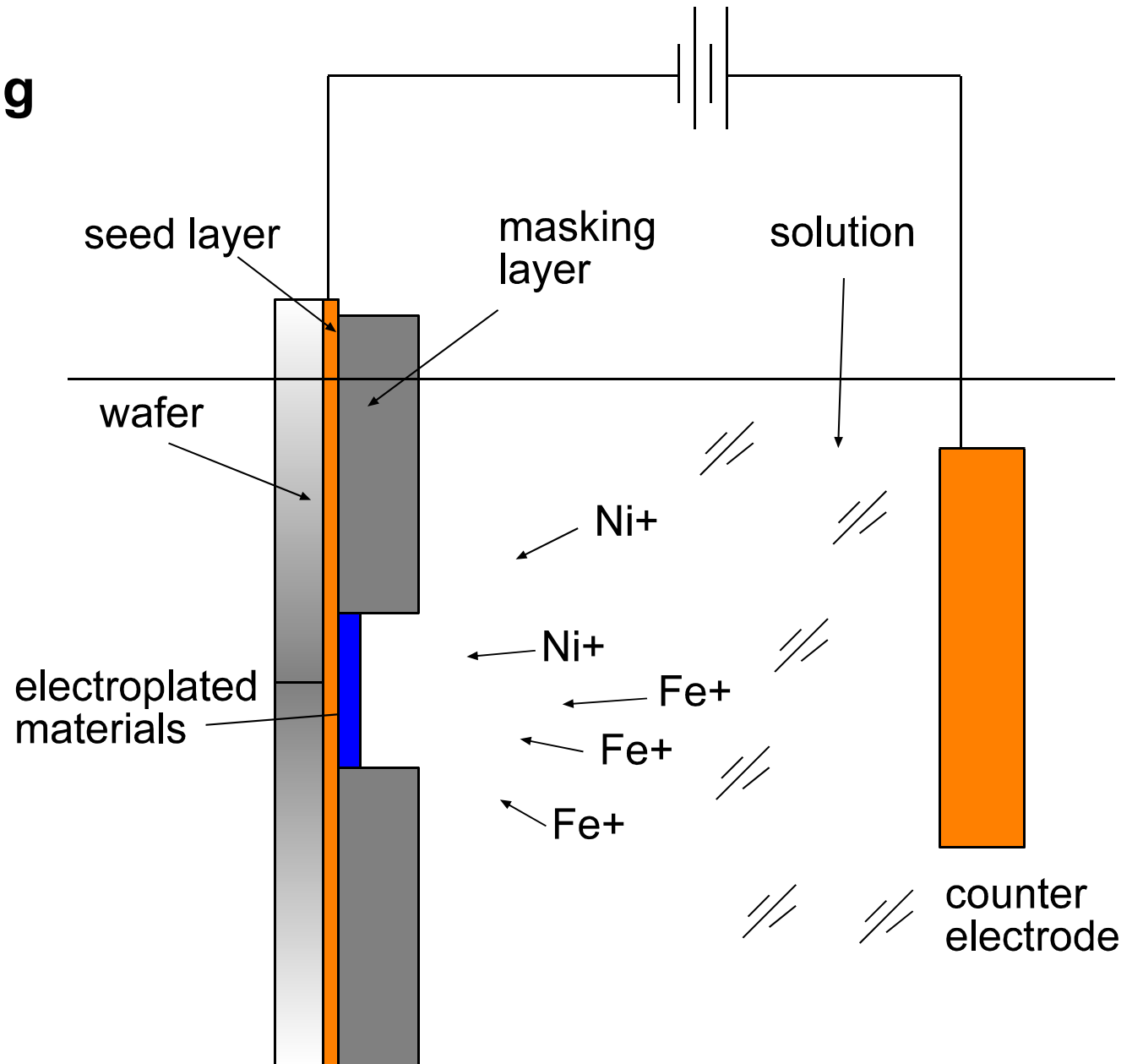


RF
plasma
generator



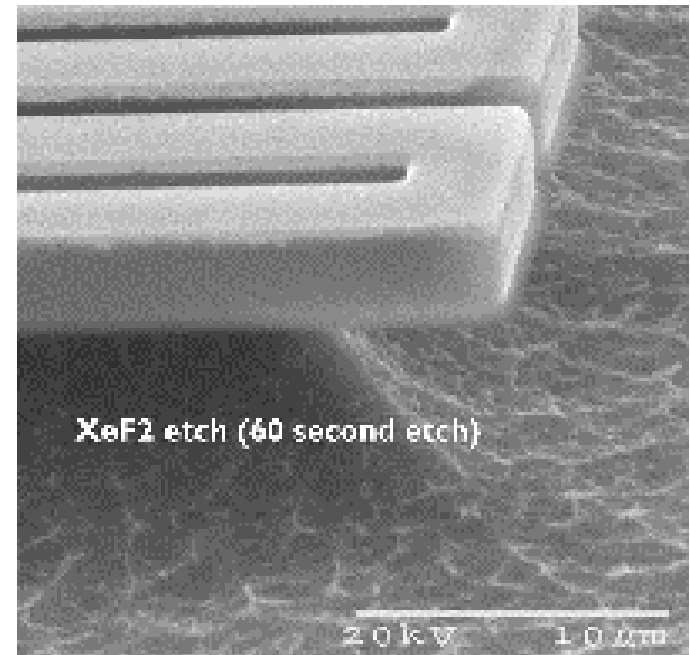
Electroplating

- Electroplating process description



Gas Phase Silicon Etching

- XeF_2
 - liquid phase under room temperature
 - $2\text{XeF}_2 + \text{Si} \Rightarrow 2\text{Xe} + \text{SiF}_4$
 - vapor phase under low pressure
 - etches silicon with high speed
 - <http://www.xactix.com/>
- BrF_3
 - solid phase under regular pressure and room temperature
 - vapor phase (sublimation) under low pressure
 - BrF_3 when reacted with water turns into HF at room temperature.
- Both are isotropic etchants



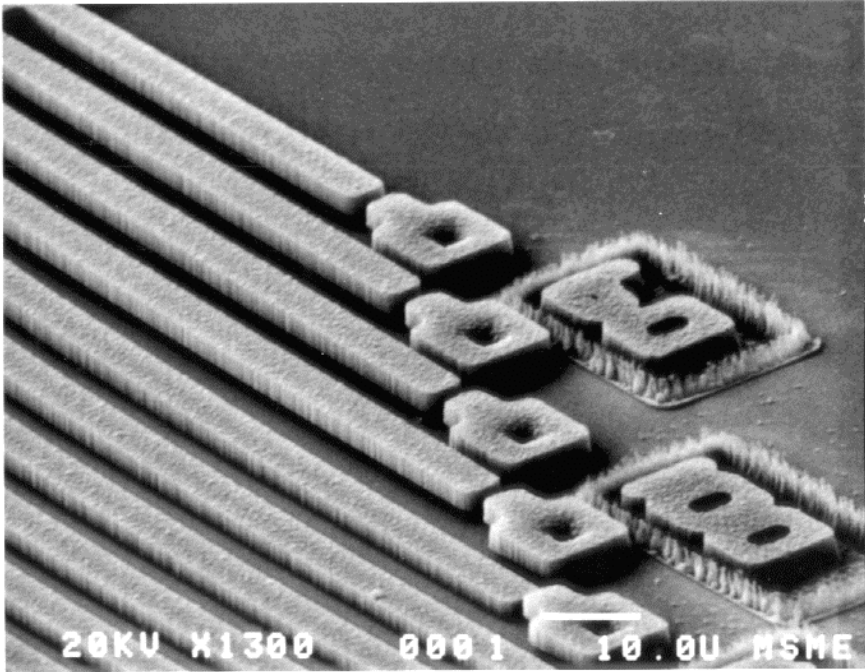
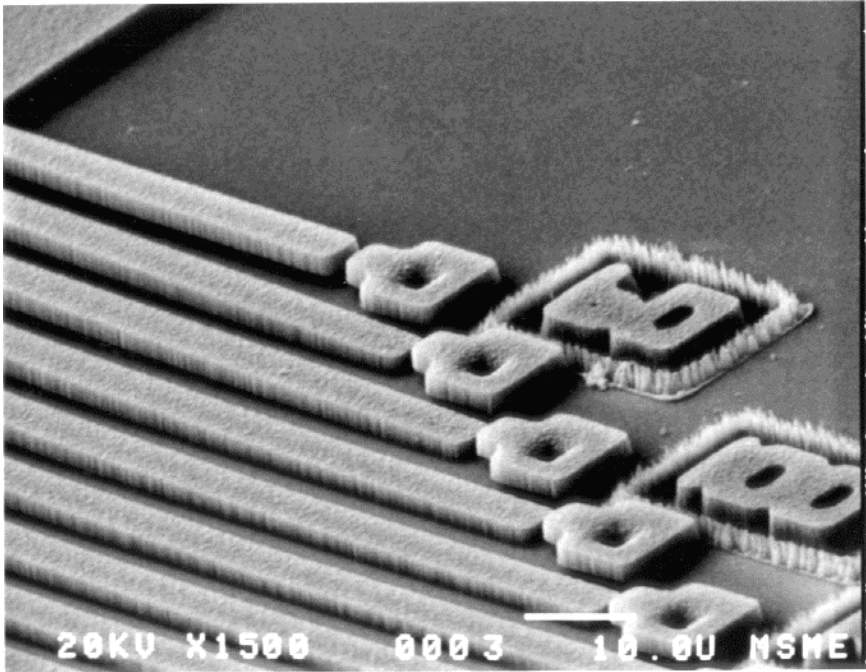
Organic Sacrificial Layer

- Photoresist
 - etching by plasma etching (limited lateral etch extent)
 - or by organic solvents (acetone or alcohol)
- Polyimide
 - etching by organic solvents
- Advantage
 - extremely low temperature process
 - easy to find structural solutions with good selectivity
- Disadvantage
 - many structural layers such as LPCVD are not compatible.
 - Metal evaporation is also associated with high temperature metal particles, so it is not completely compatible and caution must be used.

Criteria for Selecting Materials and Etching Solutions

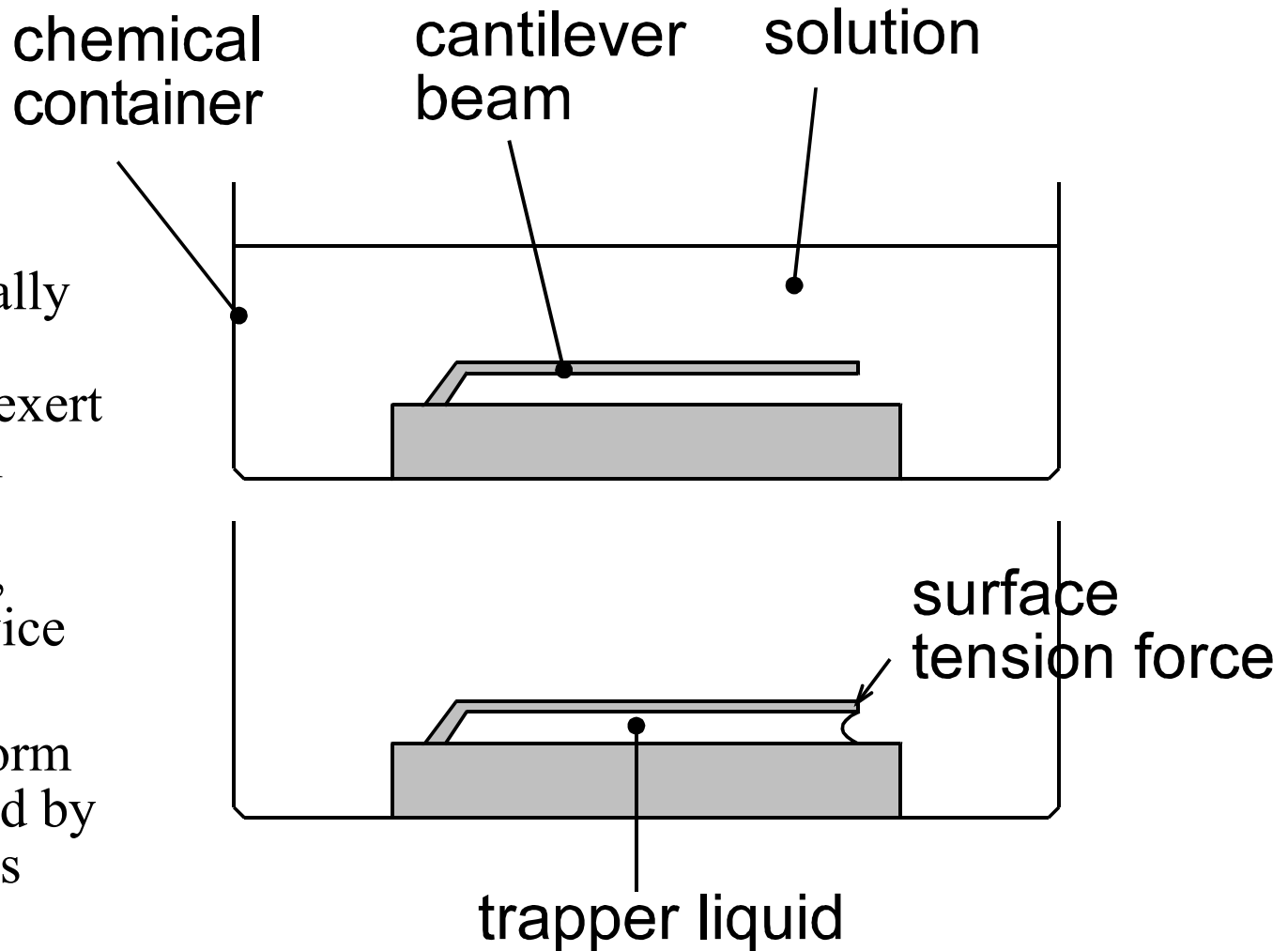
- **Selectivity**
 - etch rate on structural layer/etch rate on sacrificial layer must be high.
- **Etch rate**
 - rapid etching rate on sacrificial layer to reduce etching time
- **Deposition temperature**
 - in certain applications, it is required that the overall processing temperature be low (e.g. integration with CMOS, integration with biological materials)
- **Intrinsic stress of structural layer**
 - to remain flat after release, the structural layer must have low stress
- **Surface smoothness**
 - important for optical applications
- **Long term stability**

Stiction = Sticking and Friction



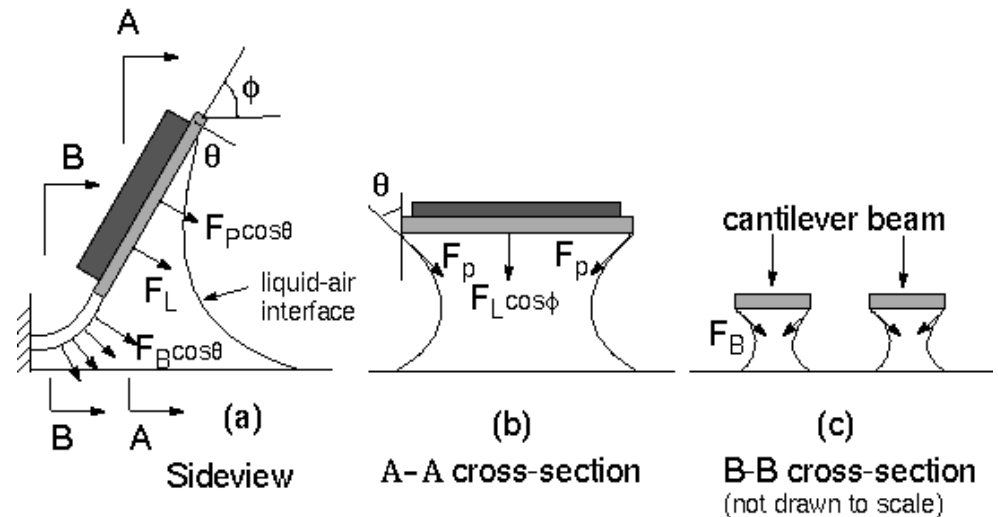
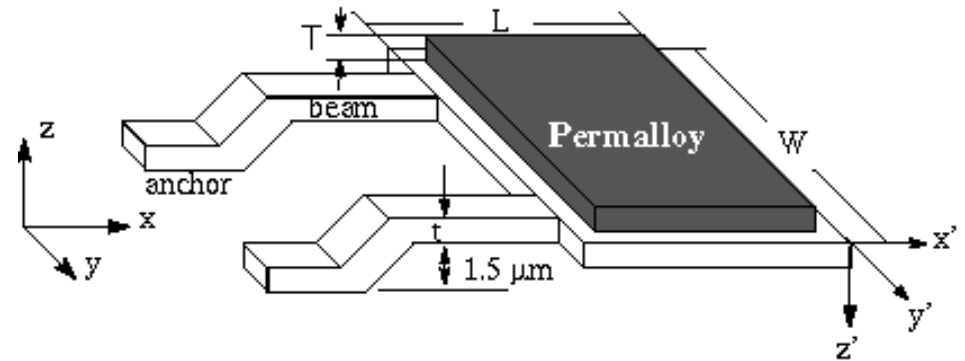
Origin of Stiction

- As the liquid solution gradually vaporizes, the trapped liquid exerts surface tension force on the microstructure, pulling the device down.
- Surfaces can form permanent bond by molecule forces when they are close.



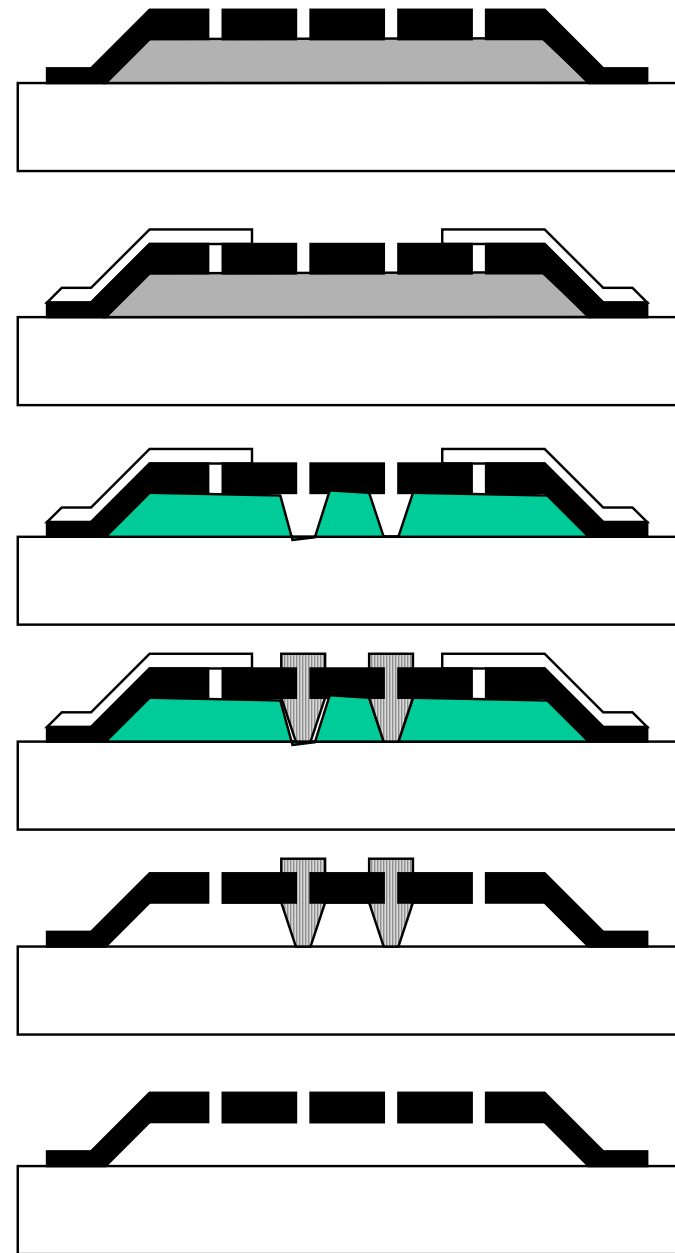
Antistiction Method I - Active Actuation Method

- Use magnetic actuation to pull structures away from the surface
 - reduced surface tension length of arm
- Limitations
 - only works for structures with magnetic material.



Antistiction Method II - Organic Pillar

- Use organic pillar to support the structure during the liquid removal.
- The organic pillar is removed by oxygen plasma etching.



Antistiction Drying Method III - Phase Change Release Method

Supercritical CO₂ Drying

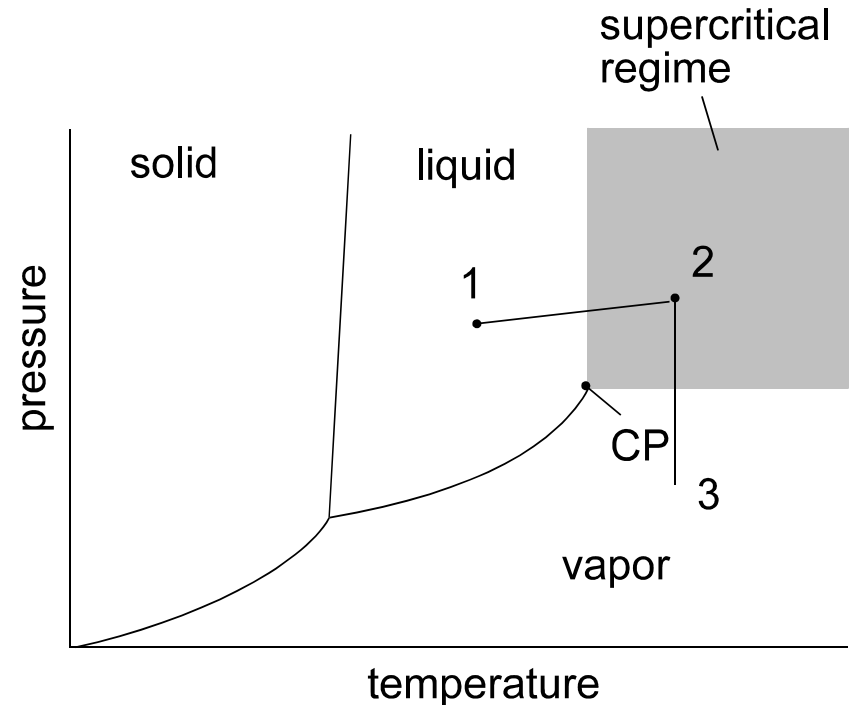
- Avoid surface tension by relying on phase change with less surface tension than water-vapor.
- * p. 128-129
- Supercritical state: temp > 31.1 °C and pressure > 72.8 atm.
- Step 1: change water with methanol
- Step 2: change methanol with liquid carbon dioxide (room temperature and 1200 psi)
- Step 3: content heated to 35 °C and the carbon dioxide is vented.

- Free-standing cantilever beams upto 850 μm can stay released.



Super Critical Drying

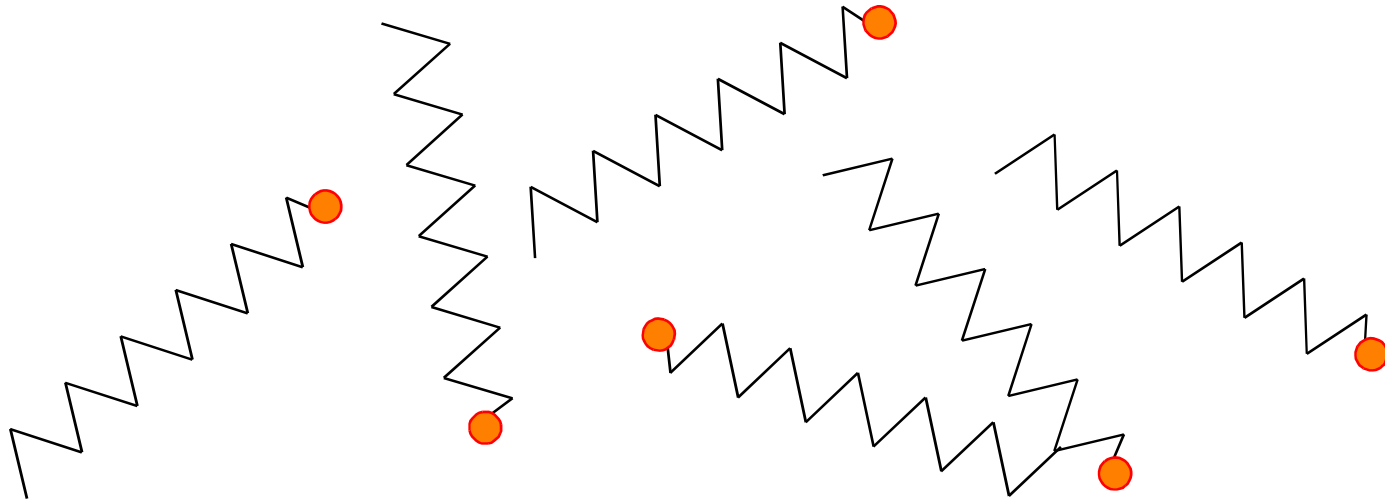
- When a substance in the liquid phase at a pressure greater than the critical pressure is heated, it undergoes a transition from a liquid to a supercritical fluid at the critical temperature.
- This transition does not involve interfaces.
- Criteris
 - chemically inert, non-toxic
 - low critical temperature
- CO₂
 - critical temperature 31.1 °C
 - critical pressure 72.8 atm.(or 1073 psi)



- Exchange methanol with liquid CO₂ at 25°C and 1200 psi
- closeoff vessel and heated to 35 °C, no interface is formed.
- Vent vessel at a constant temperature above critical temperature.

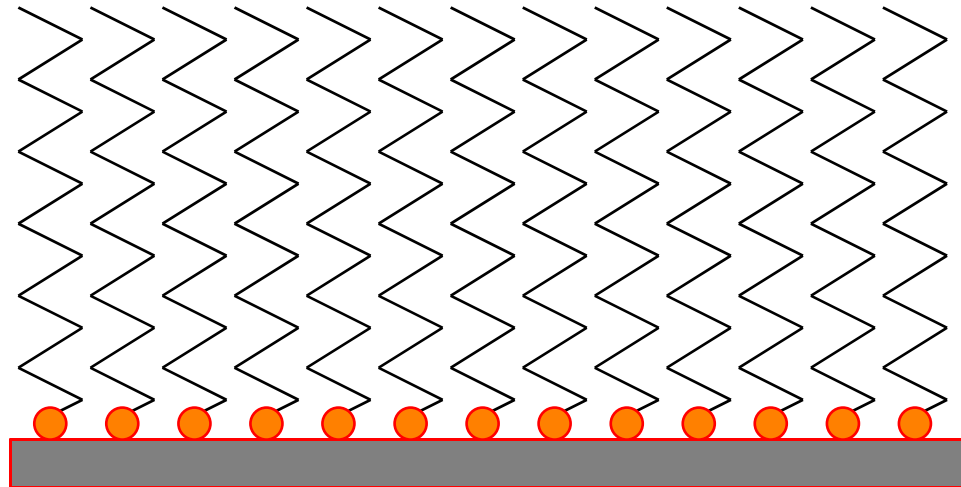
Antistiction Method III - Self-assembled Monolayer

- Forming low stiction, chemically stable surface coating using self-assembly monolayer (SAM)
- SAM file is comprised of close packed array of alkyl chains which spontaneously form on oxidized silicon surface, and can remain stable after 18 months in air.
- OTS: octadecyltrichlorosilane (forming $C_{18}H_{37}SiCl_3$)

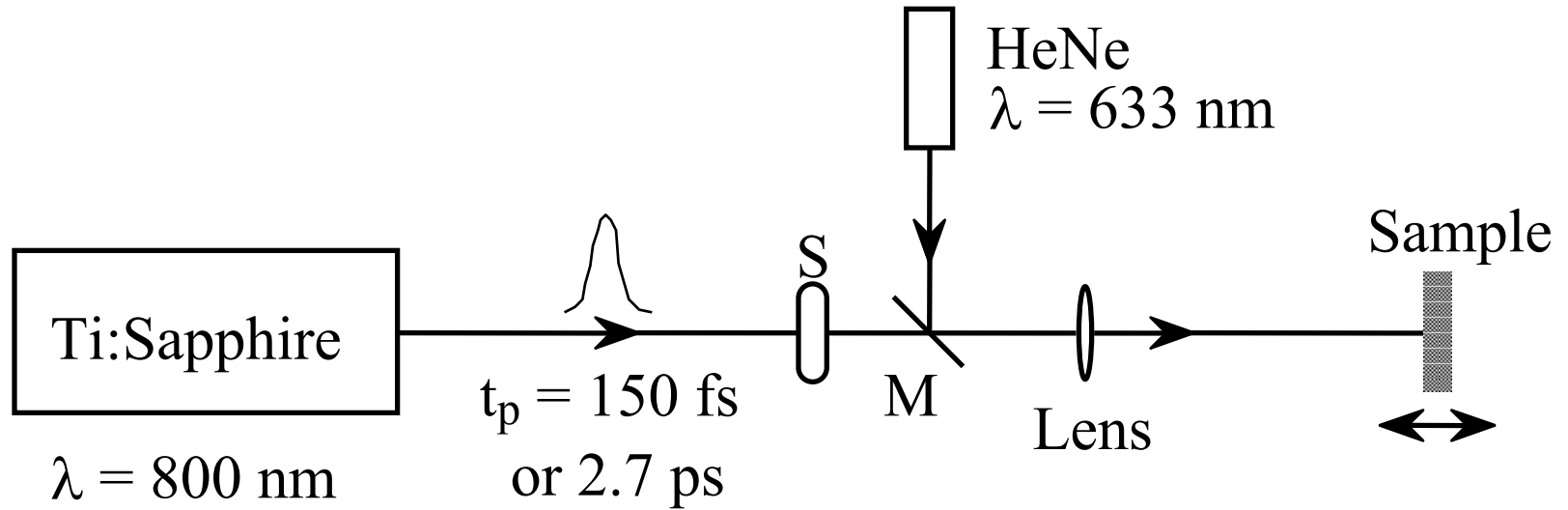


Result of SAM Assembly

- Surface oxidation: H₂O₂ soak
- SAM formation
 - isopropanol alcohol rinse
 - CCl₄ rinse
 - OTS solution
 - CCl₄ rinse



Short Pulse Laser Irradiation



Undoped Polysilicon

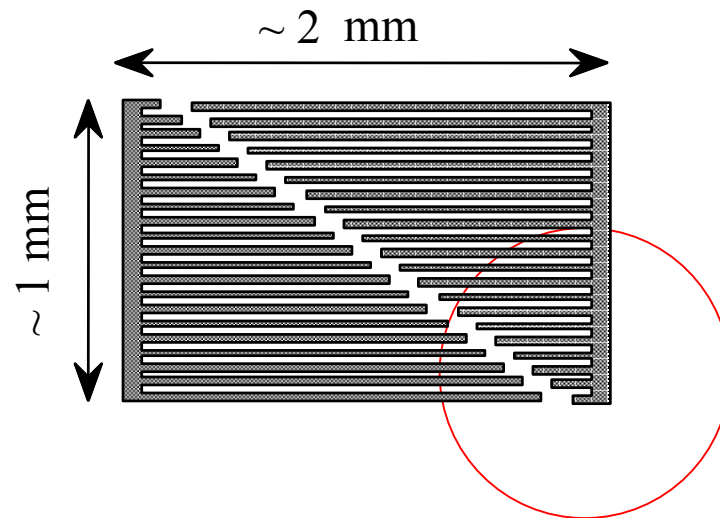
Cantilever Array

length: $60 \mu\text{m}$ to 1 mm

width: $5 \mu\text{m}$

thickness: $2 \mu\text{m}$

gap height: $1.5 \mu\text{m}$



Faculty: Phinney