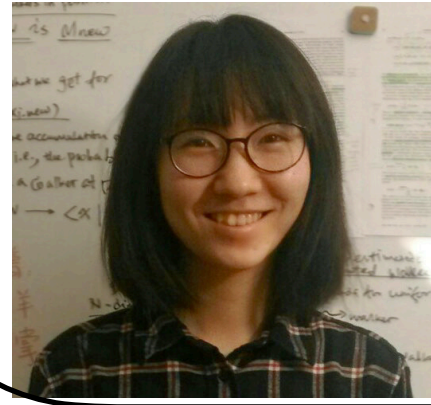
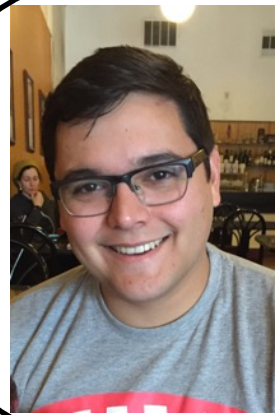
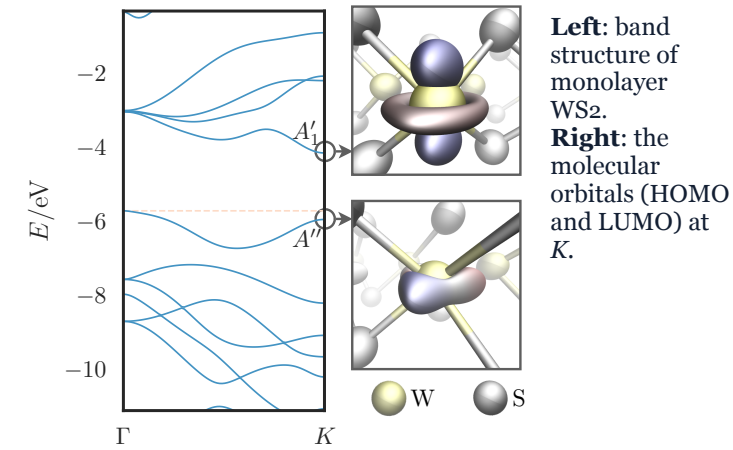


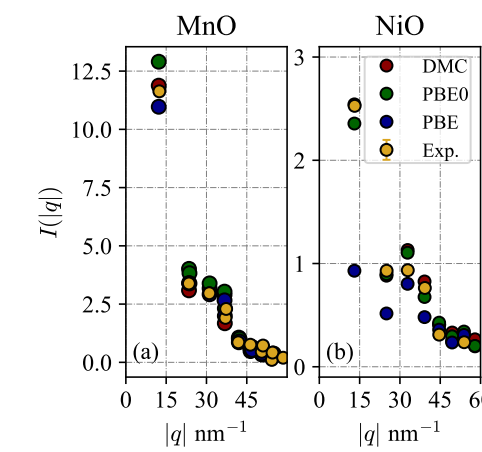
Poster advertisements



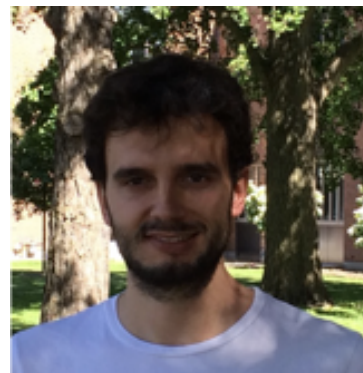
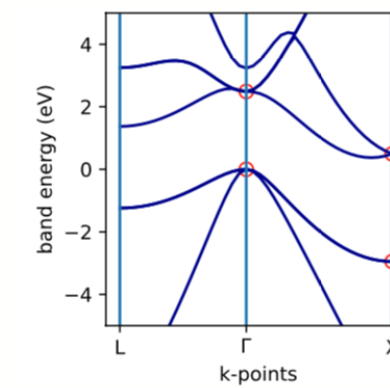
Practical spin-orbit in QMC
Yueqing Chang



Neutron intensities using QMC
Alex Munoz



Band structures using QMC
William Wheeler, Shivesh Pathak



Electron-spin coupling as a predictor for “interesting” physics
Joao Rodrigues



A new prediction for spin excitations in MgTi₂O₄
Brian Busemeyer

Benchmarking high accuracy methods on realistic systems



Collaboration on the many-electron problem
Presented by Lucas K. Wagner

Kiel T. Williams,¹ Li Chen,¹ Hao Shi,^{2,3} Mario Motta,⁴ Chunyao Niu,^{3,5} Ushnish Ray,⁴ Sheng Guo,⁴ Robert J. Anderson,⁶ Jia Li,⁷ Junhao Li,⁸ Lan Nguyen Tran,^{7,9} Chia-Nan Yeh,⁷ Bastien Mussard,¹⁰ Sandeep Sharma,¹⁰ Yuan Yao,⁸ Mark van Schilfgaarde,⁶ George H. Booth,⁶ Garnet Chan,⁴ Shiwei Zhang,^{2,3} Emanuel Gull,^{7,2} Dominika Zgid,^{7,9,2} Andrew Millis,^{2,11} Cyrus Umrigar,⁴ and Lucas K. Wagner¹



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² *Center for Computational Quantum Physics, Flatiron Institute, New York, NY 10010*

³ *Department of Physics, College of William and Mary, Williamsburg, VA 23185*

⁴ *California Institute of Technology, Pasadena, CA 91125*

⁵ *School of Physics and Engineering, Zhengzhou University, Zhengzhou 450001, China*

⁶ *Department of Physics, King's College London, Strand, London, WC2R 2LS, U.K*

⁷ *Department of Physics, University of Michigan, Ann Arbor, MI 48109*

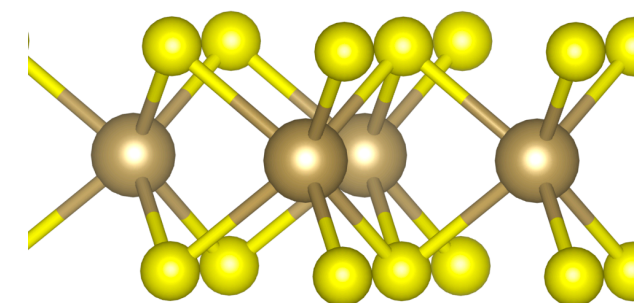
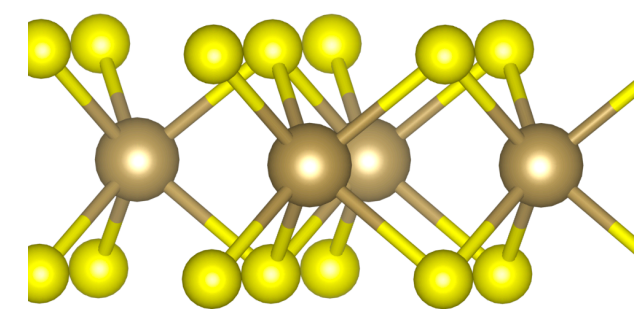
⁸ *Laboratory of Atomic and Solid State Physics, Cornell University, Ithaca, NY 14853*

⁹ *Department of Chemistry, University of Michigan, Ann Arbor, MI 48104*

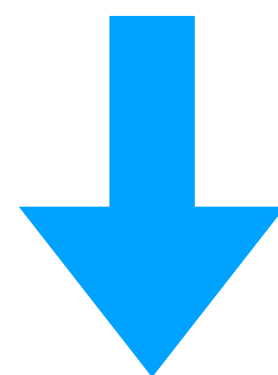
¹⁰ *Department of Chemistry, University of Colorado, Boulder*

¹¹ *Department of Physics, Columbia University, New York, NY 10027*

How reliable are first principles calculations?

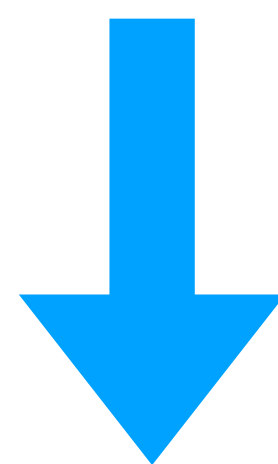


$$+ H|\Psi\rangle = i\hbar\frac{\partial}{\partial t}|\Psi\rangle$$



Approximation!

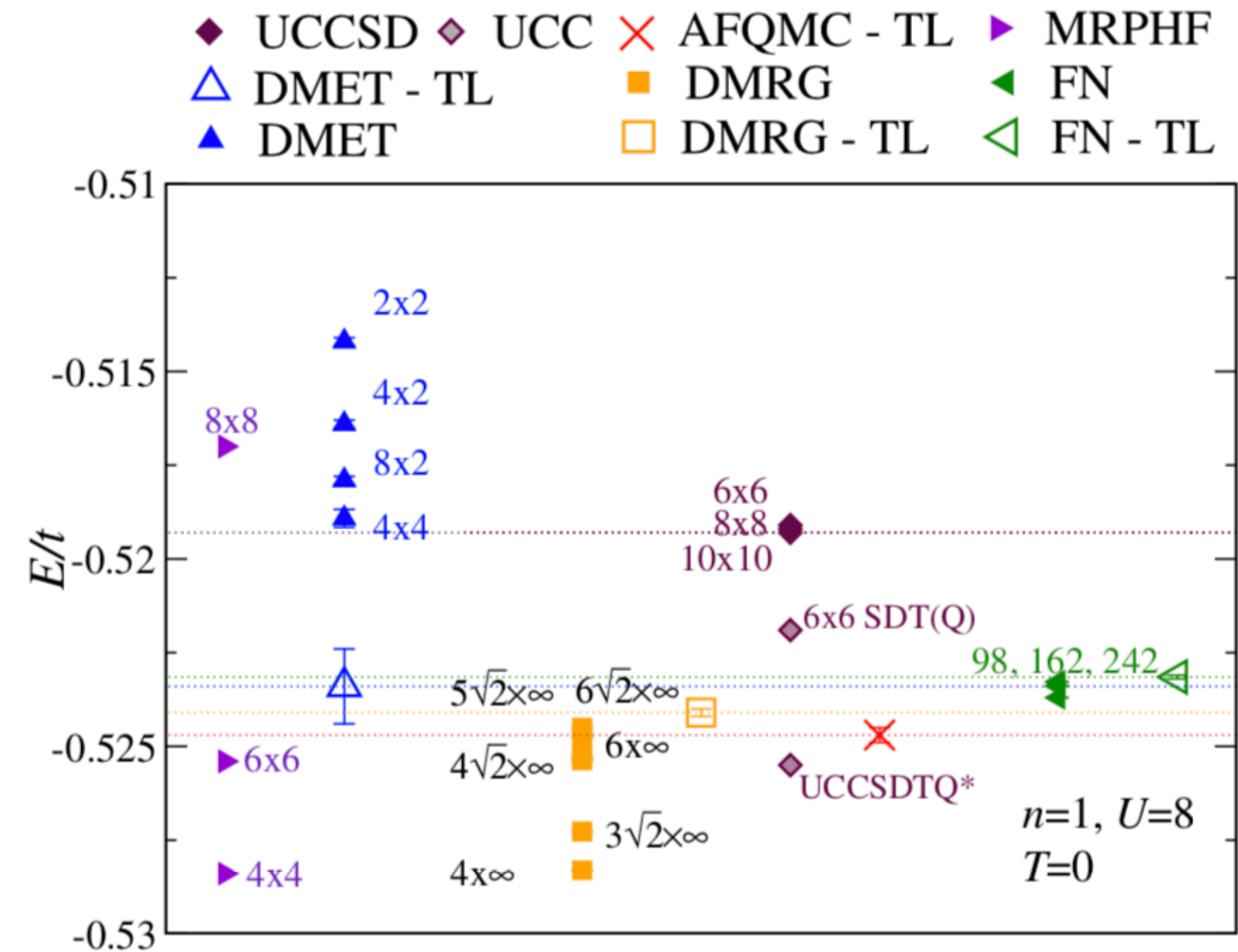
DFT, DMRG, QMC, GW, etc



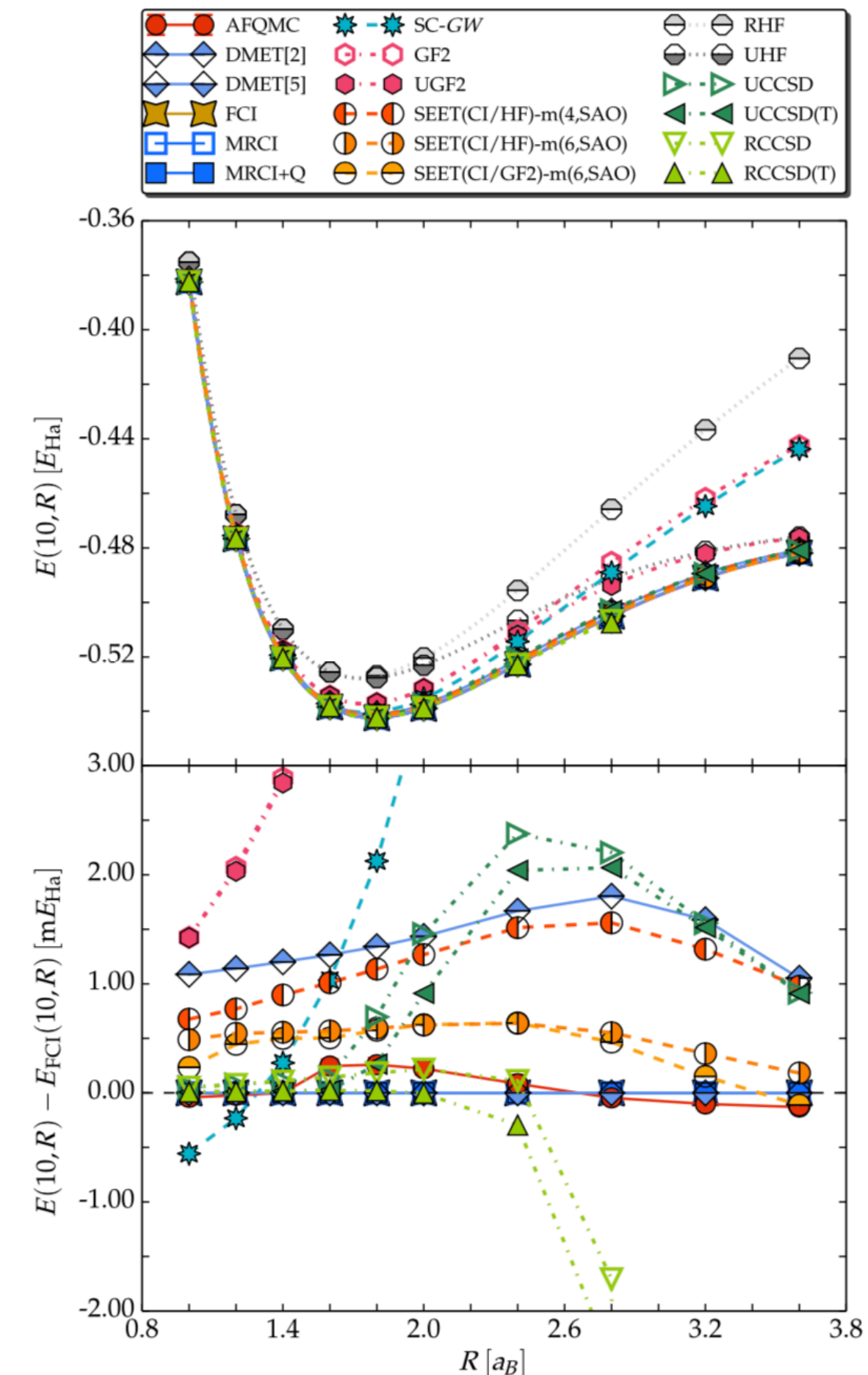
Total energy, excitations, properties

How close are these to exact solutions?

Previous work from the Collaboration



LeBlanc et al.
 Agreement between high accuracy methods
 on the Hubbard model



Motta et al.
 Chain of hydrogen atoms

Our objective

Provide total energy benchmarks for **realistic systems** on well-defined Hamiltonians.

Important systems: **transition metal compounds**, which exhibit strongly correlated behavior in materials

Simplify: **ground state energy**.

Choosing a Hamiltonian

Frozen core

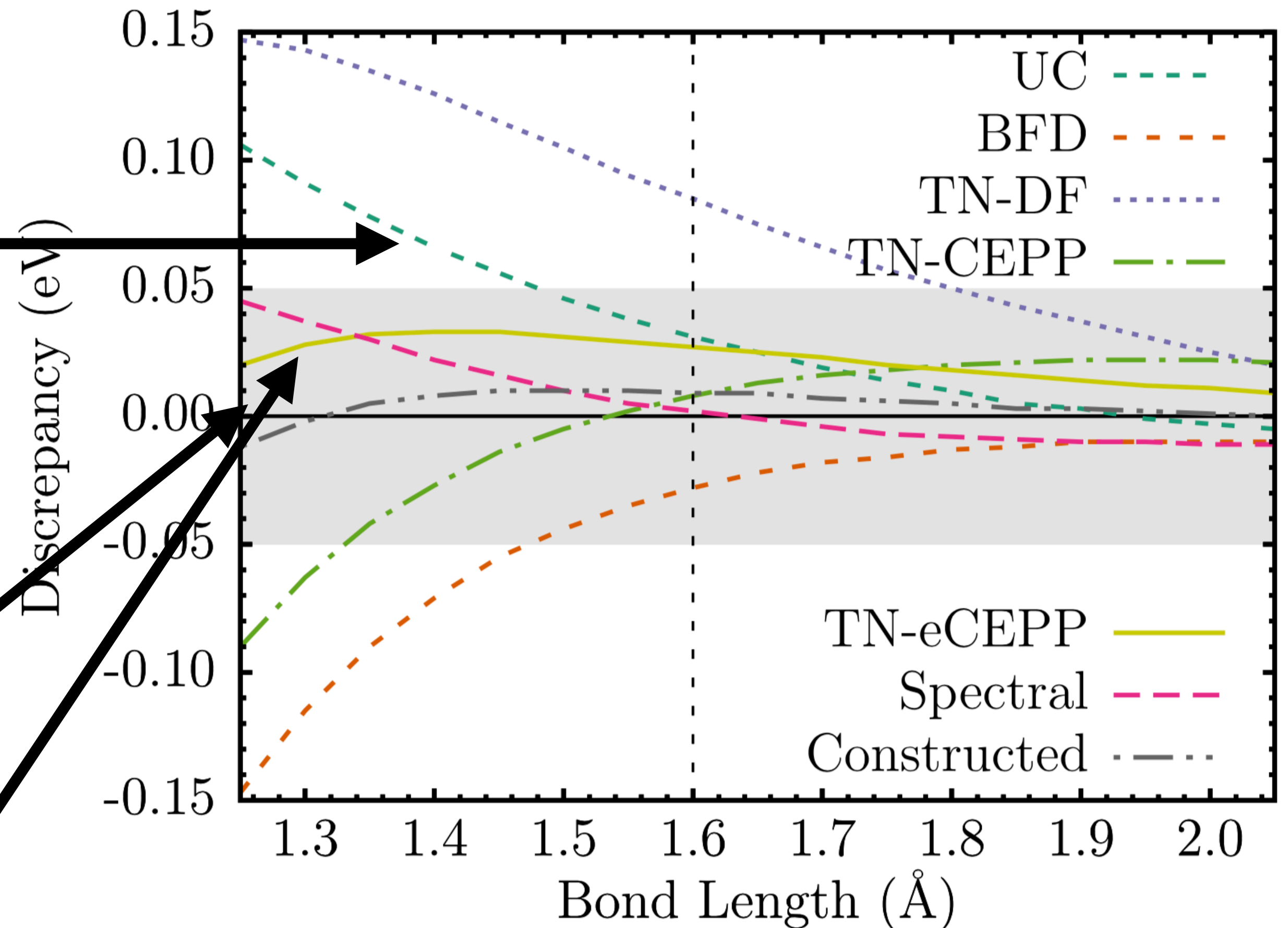
- Complex to specify
- Not necessarily more accurate than ECPs

All electron

- Very expensive
- DMC is hard

ECPs (Trail and Needs)

- Everyone can do it
- Accurate and available



Bennett, Melton, Annaberdiyev, Wang,
Shulenberger, Mitas
J. Chem. Phys. **147**, 224106 (2017)

Finite basis sets

$$\hat{H} = -\frac{1}{2} \sum_i \nabla_i^2 + \hat{V}$$

For a single Cr atom:

Number of one-particle orbitals

Approximate size of Hilbert space

Project onto discrete basis

$$H_{ij} = \langle i | \hat{H} | j \rangle$$

Sparse representation

$$\hat{H} = \sum_{ab} t_{ab} c_a^\dagger c_b + \sum_{abcd} V_{abcd} c_a^\dagger c_b^\dagger c_c c_d$$

vdz	54	10^{15}
vtz	88	10^{19}
vqz	135	10^{21}
v5z	197	10^{24}
cbs	infinite	infinite

Methods and (roughly) governing equations

Density functional theory

LDA, PBE, SCAN, HSE06

$$E = f[\rho]$$

Green's function

QSGW, SC-GW, GF2, RPA

$$G = G_0 + G_0 \Sigma G$$

Many-body wave function

AFQMC, CCSD, CCSD(T), CIC+Q, CISD,
DMC, DMRG, FCIQMC, MRLCC, SHCI

$$\hat{H} |\Psi\rangle = E |\Psi\rangle$$

Embedding

SEET

Packages used

PySCF (checked Molpro)

Density functional theory
Coupled cluster
Configuration Interaction
DMRG

Integral generation

QWalk (checked CHAMP)

Fixed node diffusion QMC

MolGW (Bruneval)

QSGW (van Schilfgaard)

Home-grown codes

Self-consistent GW (Gull)

Semistochastic heat bath iteration (Umrigar)

Auxiliary field QMC (Zhang)

Multi-reference linearized coupled cluster
(Sharma)

Full CI QMC (Booth)

Database

Sc, Ti, V, Cr,
Mn, Fe, Cu,
0,+ ,M-O

×

vdz, vtz,
vqz, v5z

×

QC,
QMC,
GF, ...

≈

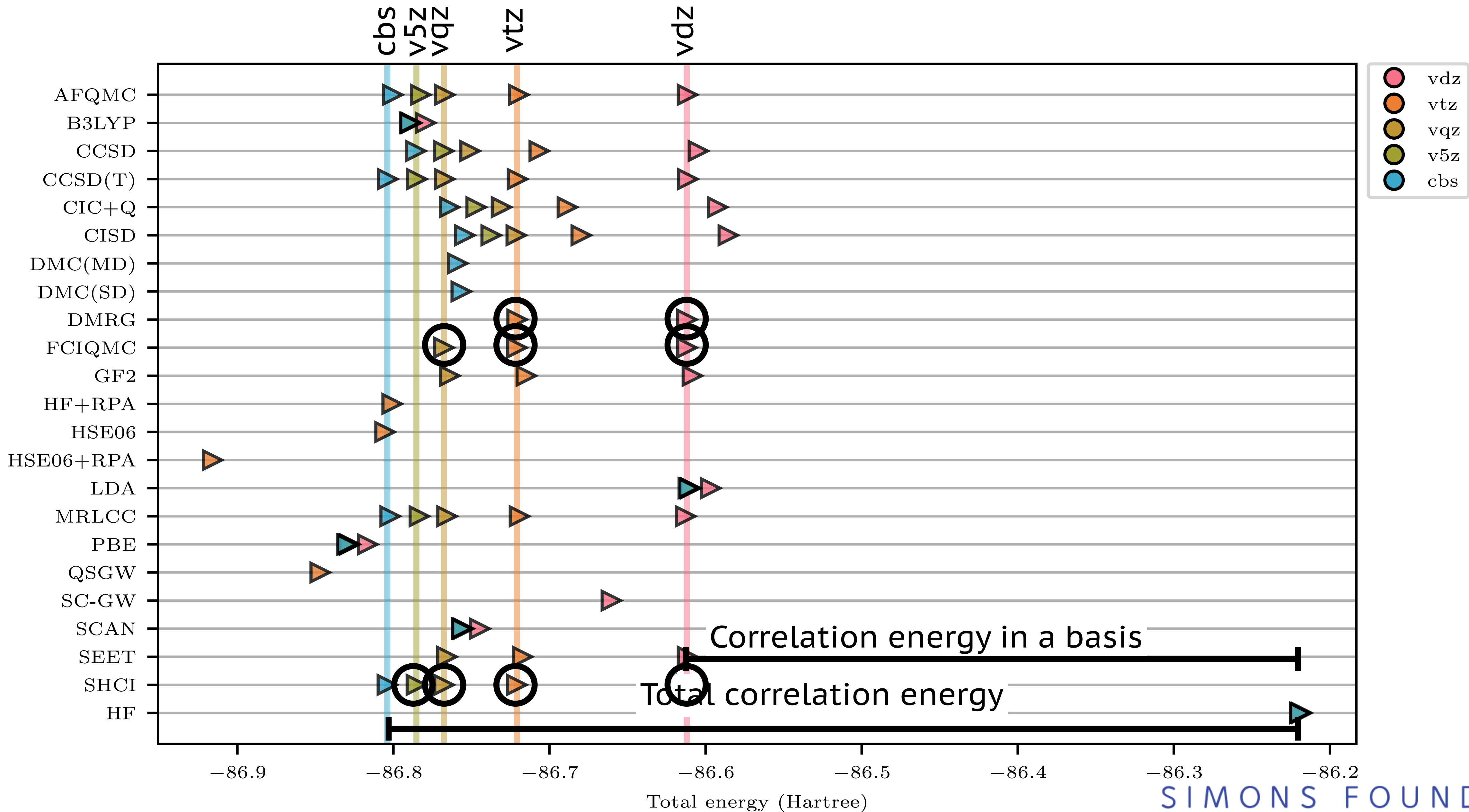
4038
entries

18 TM systems
neutral atoms,
ions and
oxides

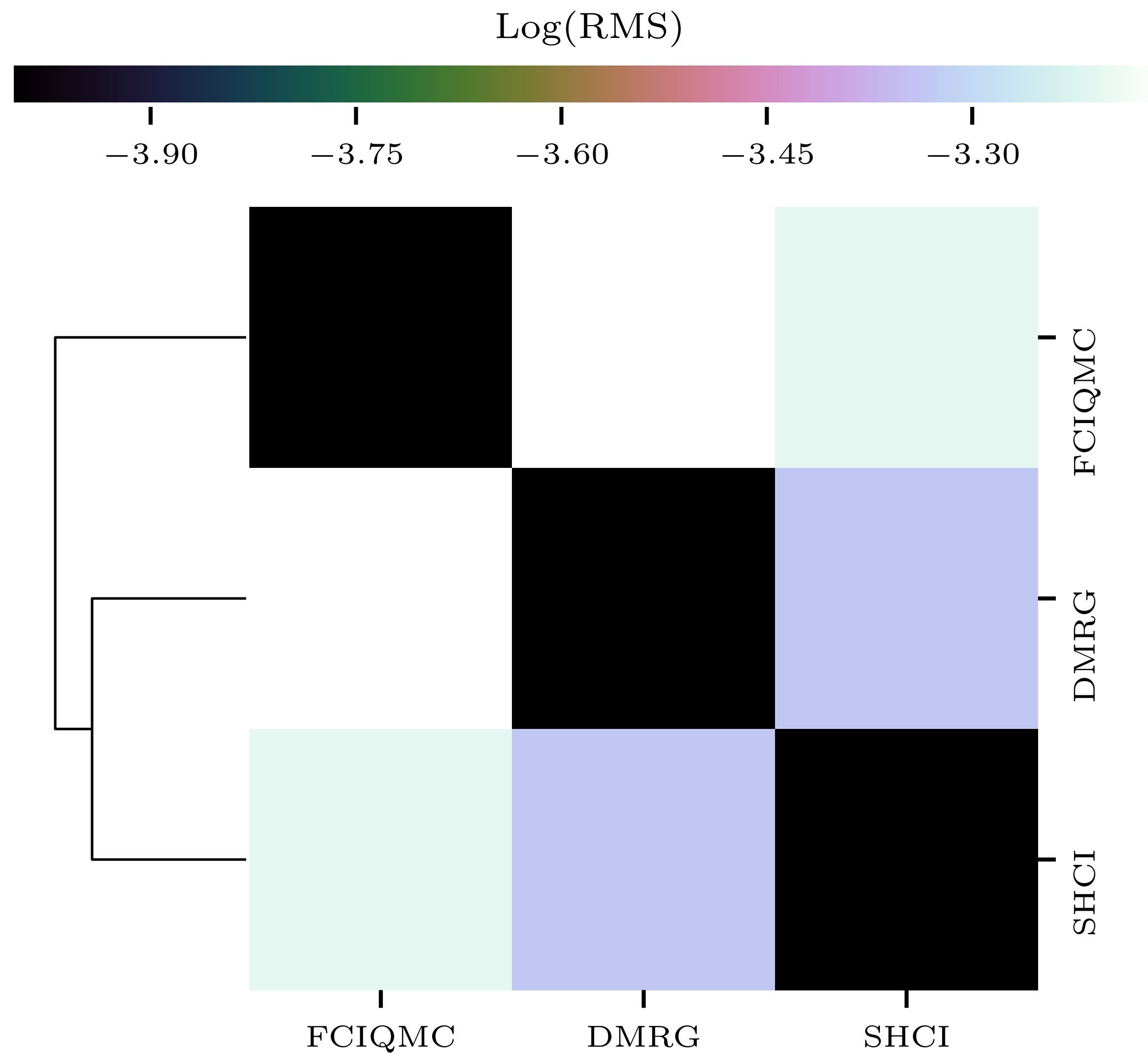
5 basis sets

20 methods

Analysis of data (Cr atom)



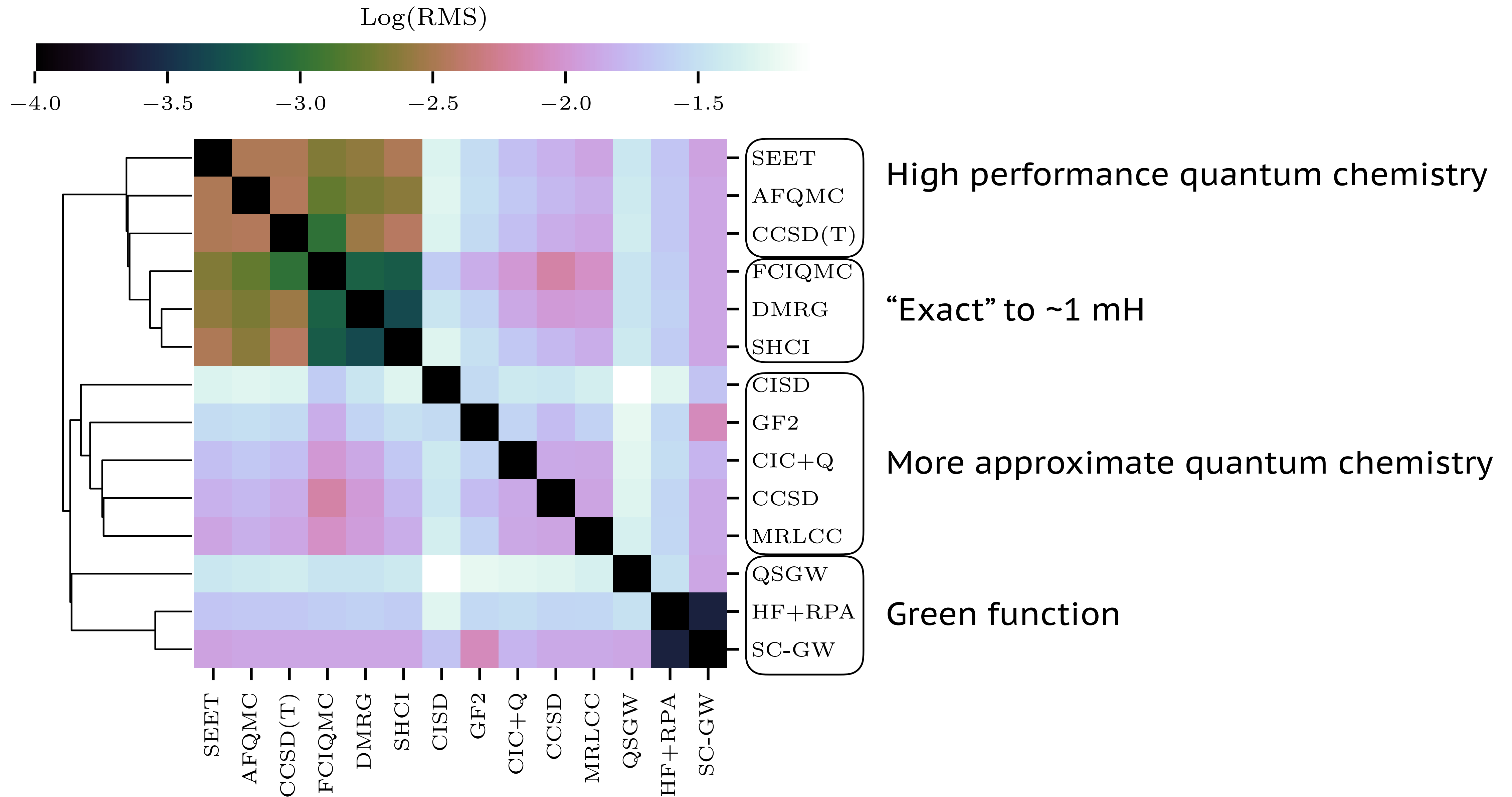
Setting a reference



3 methods that should converge to exact results **within the basis**.

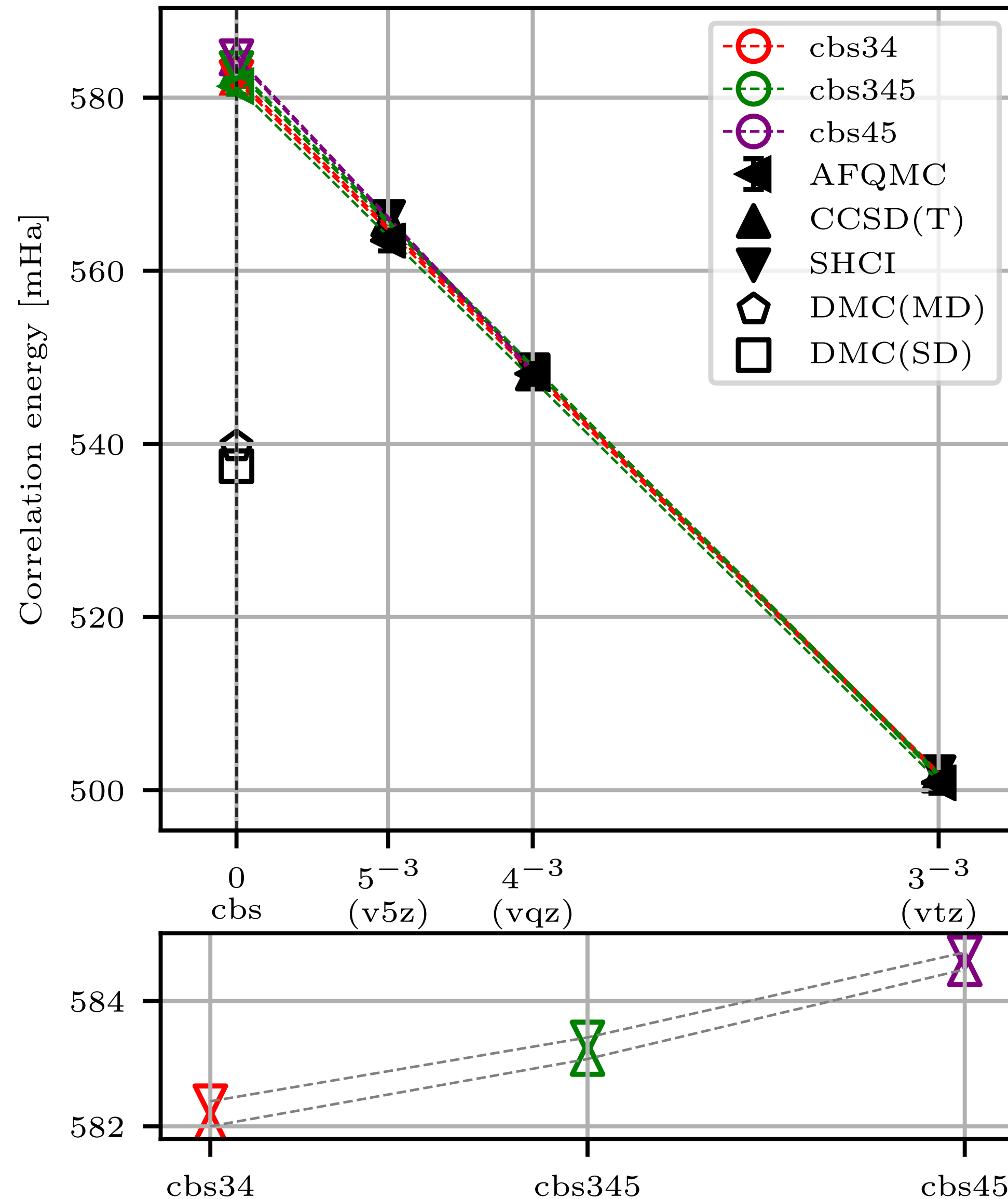
RMS deviations between 3 methods are much less than 1 mHa, around chemical accuracy.

How methods agree: cluster analysis



Basis set extrapolation: total energy

Cr atom extrapolations



Basis set error scales as $\sim 1/z^3$

Diffusion Monte Carlo uses the full Hamiltonian, but is approximate; separate lower bound on correlation energy.

Approximate uncertainty in extrapolated value of 2-5 mHa.

DFT results are compared only in the CBS limit.

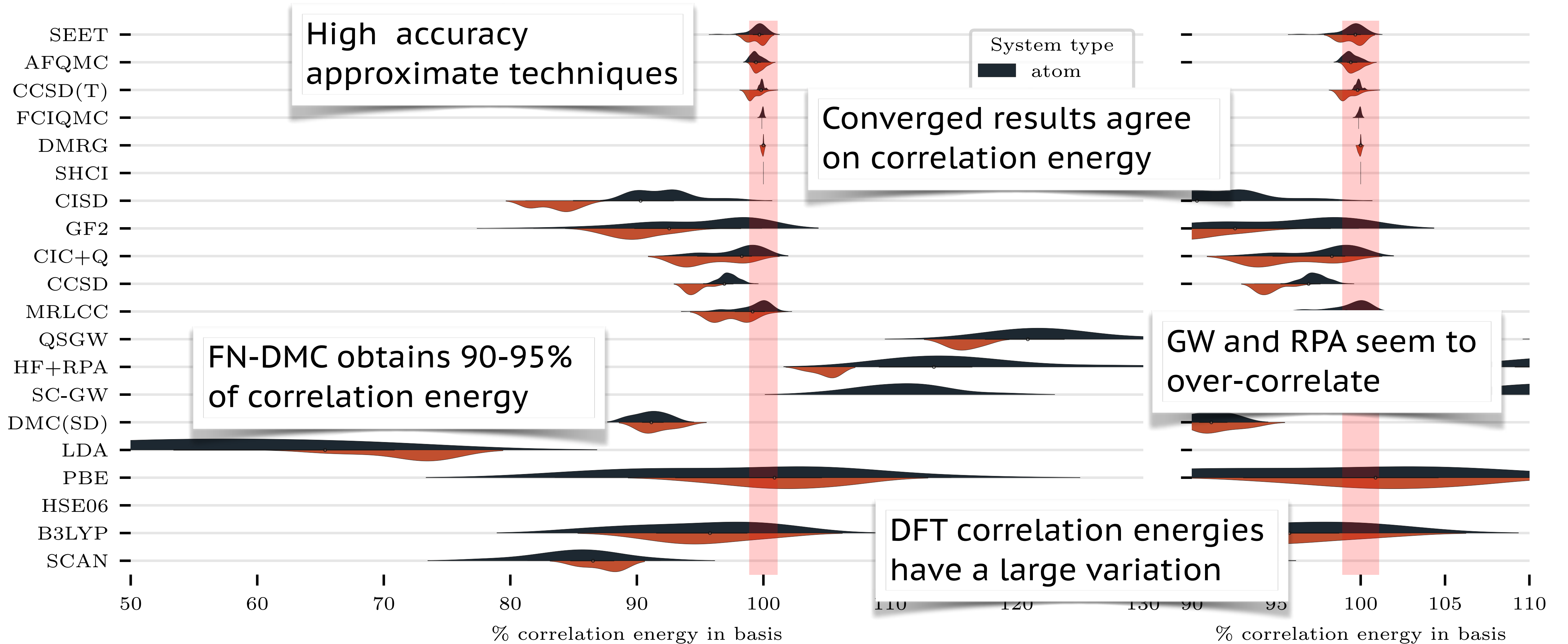
Assessing performance: correlation energy

$$100 \times \frac{E_{HF} - E_M}{E_{HF} - E_{\text{exact}}}$$

Scale-free quantity that allows us to compare different systems with different total energies

Most methods obtain consistent percentages of the correlation energy.

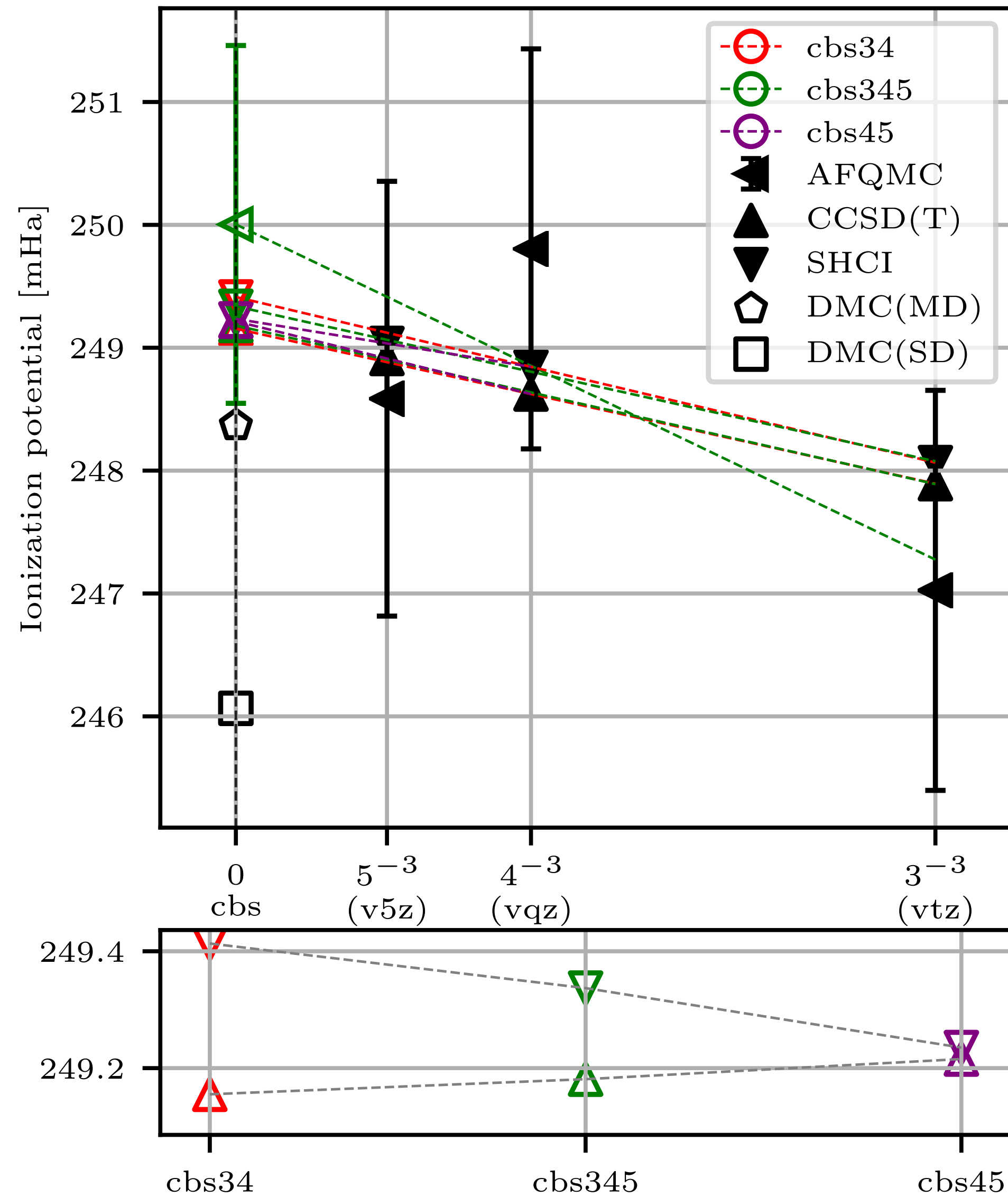
Correlation energy performance



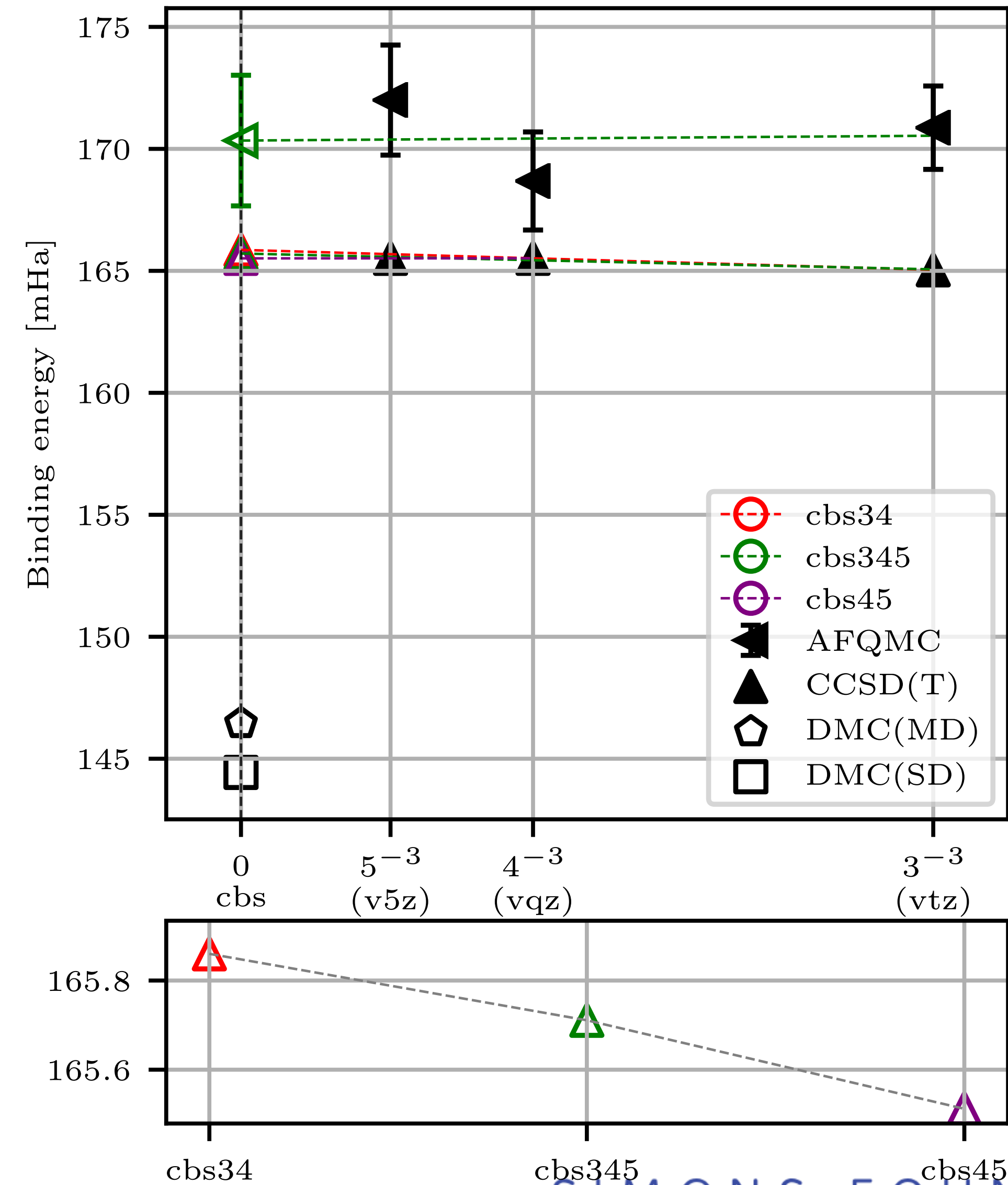
Energy differences

Basis set extrapolation: differences

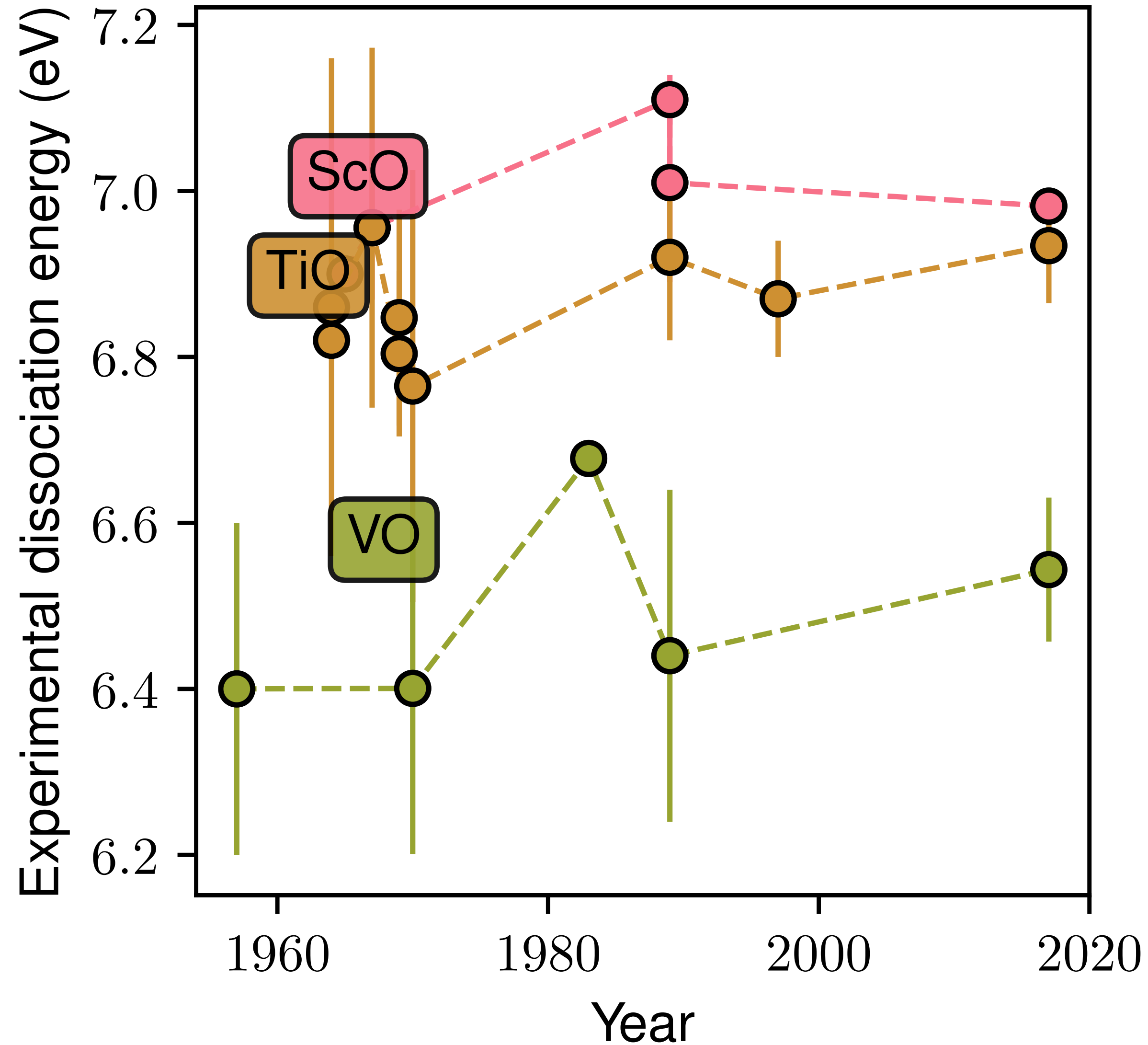
Ionization potential of Cr



Dissociation energy of CrO molecule



Energy differences: the difficulty with experiment

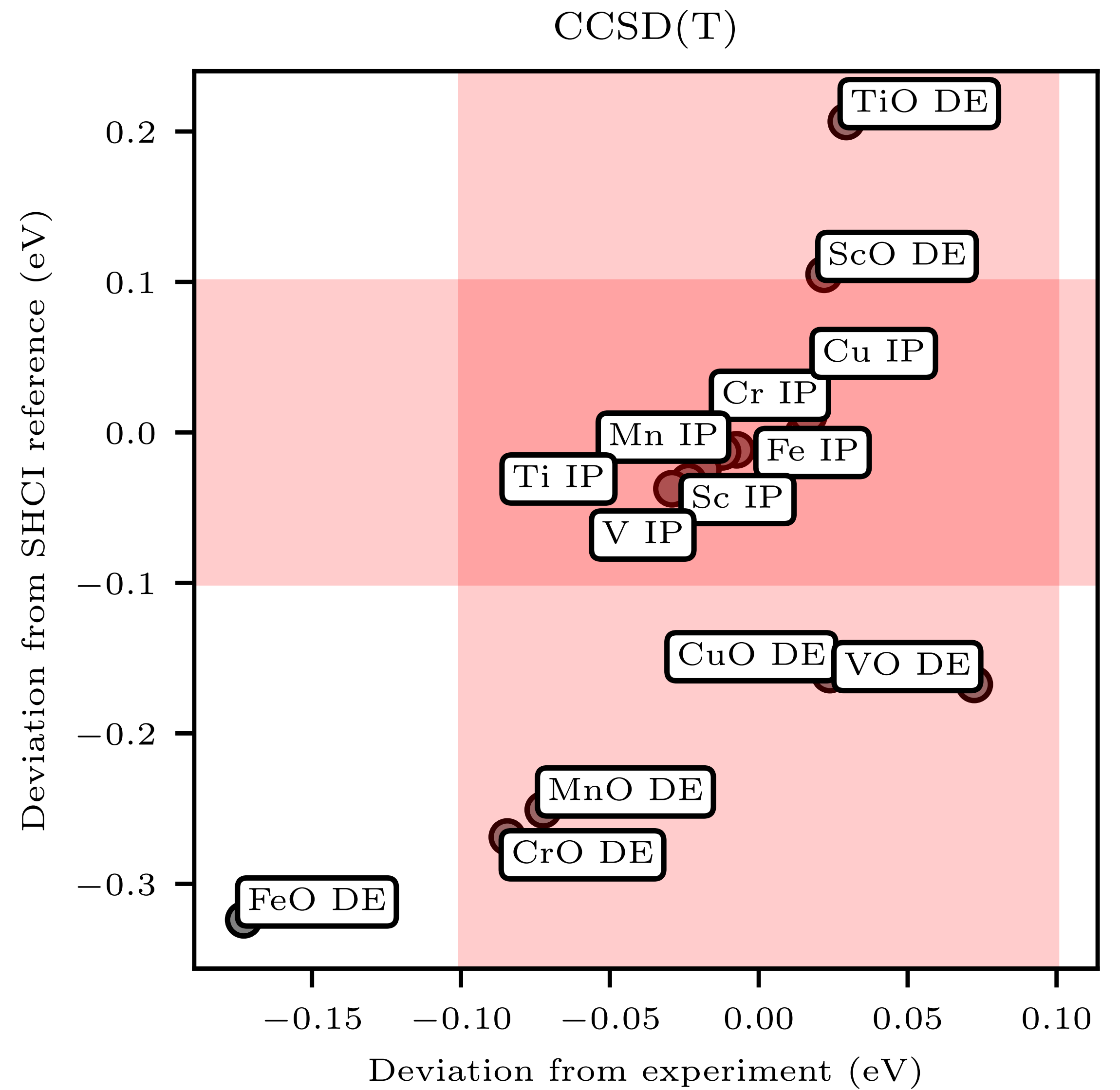
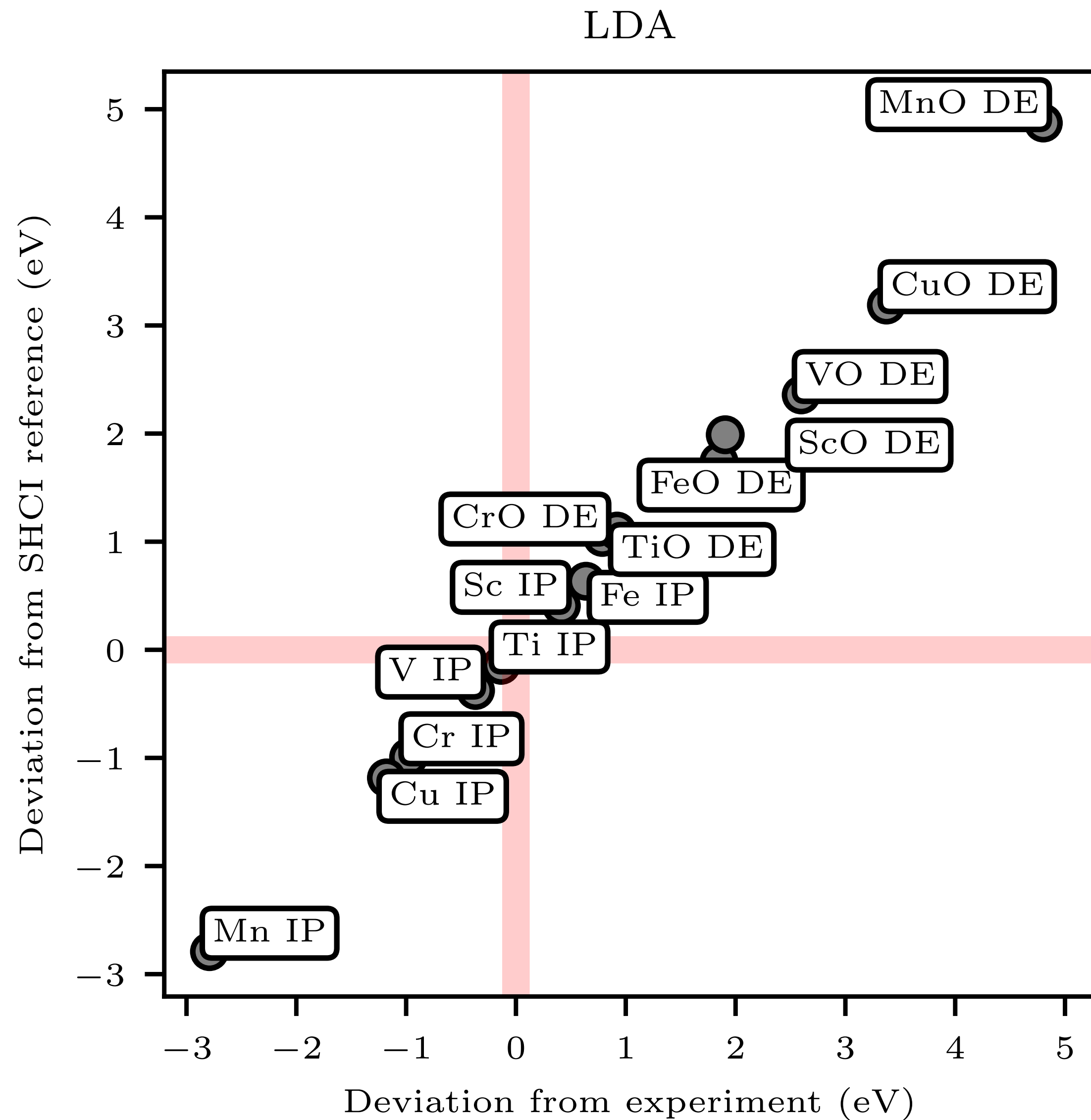


Experimental uncertainties are substantial

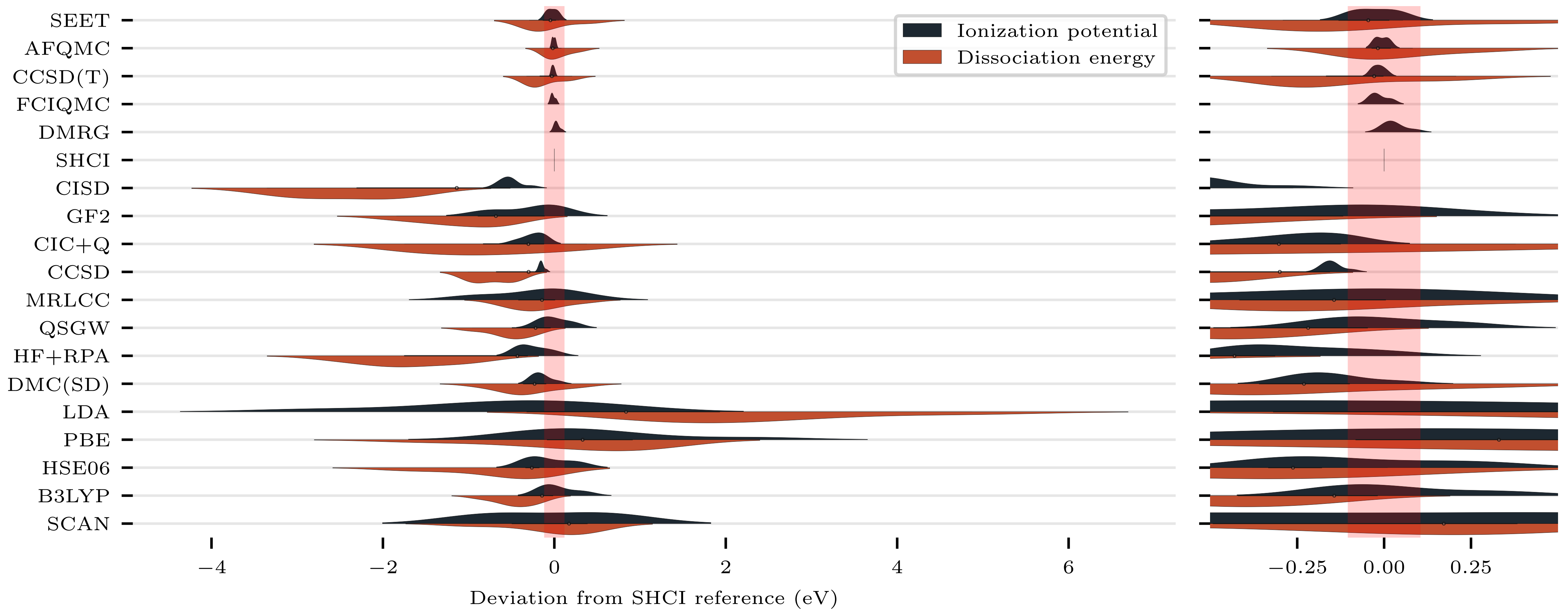
Measure the distance from the range of experimental results



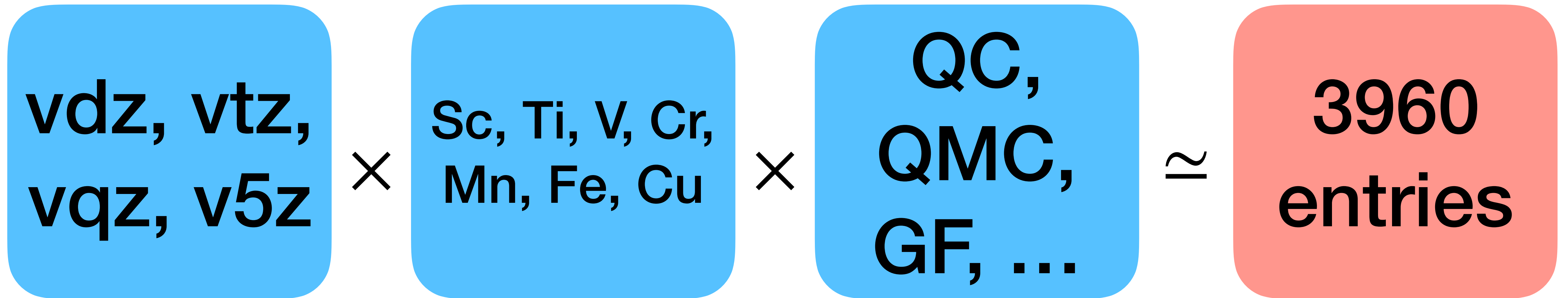
Limiting factor is experiment!



Deviation from experiment

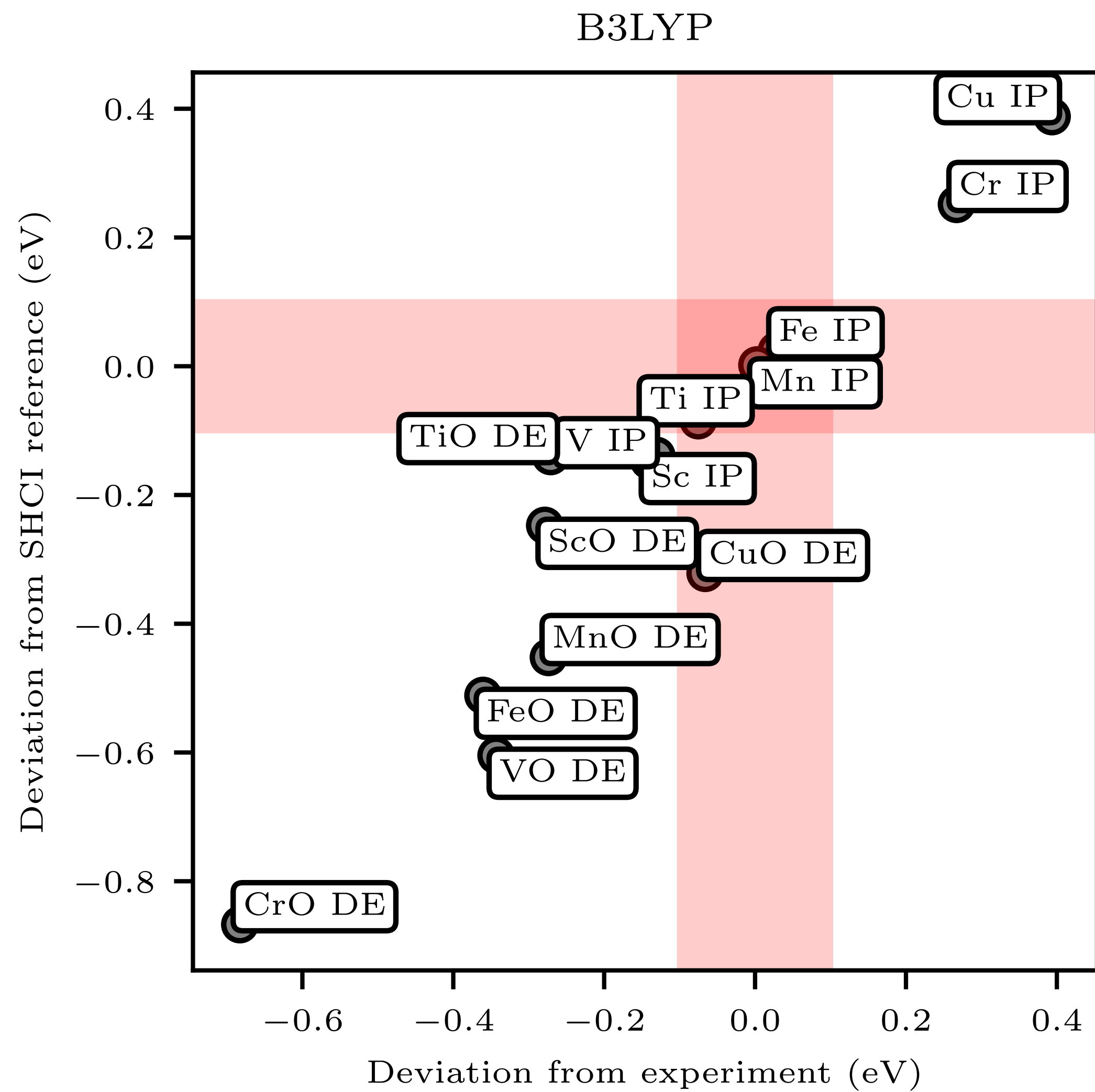
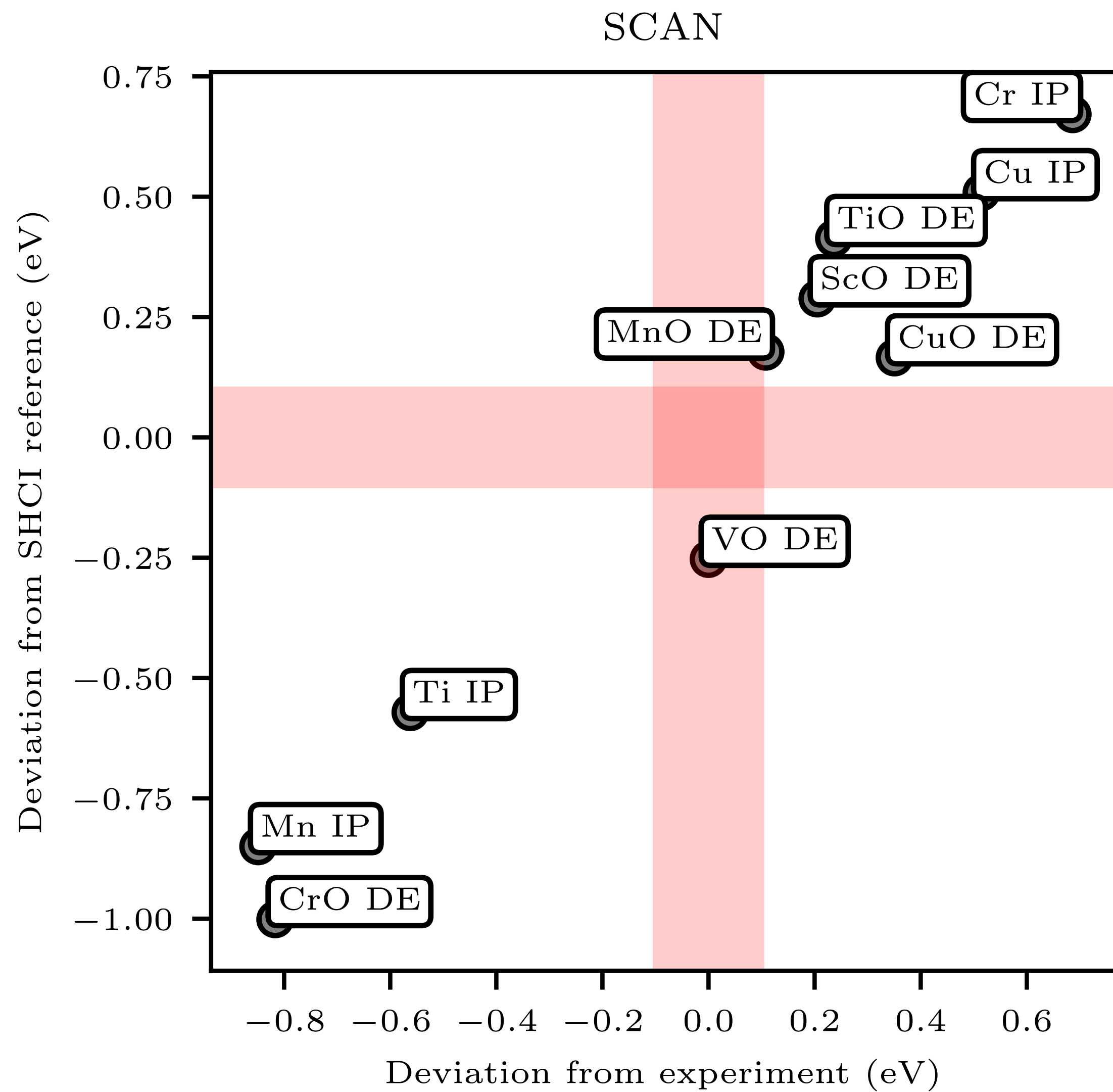


Summary

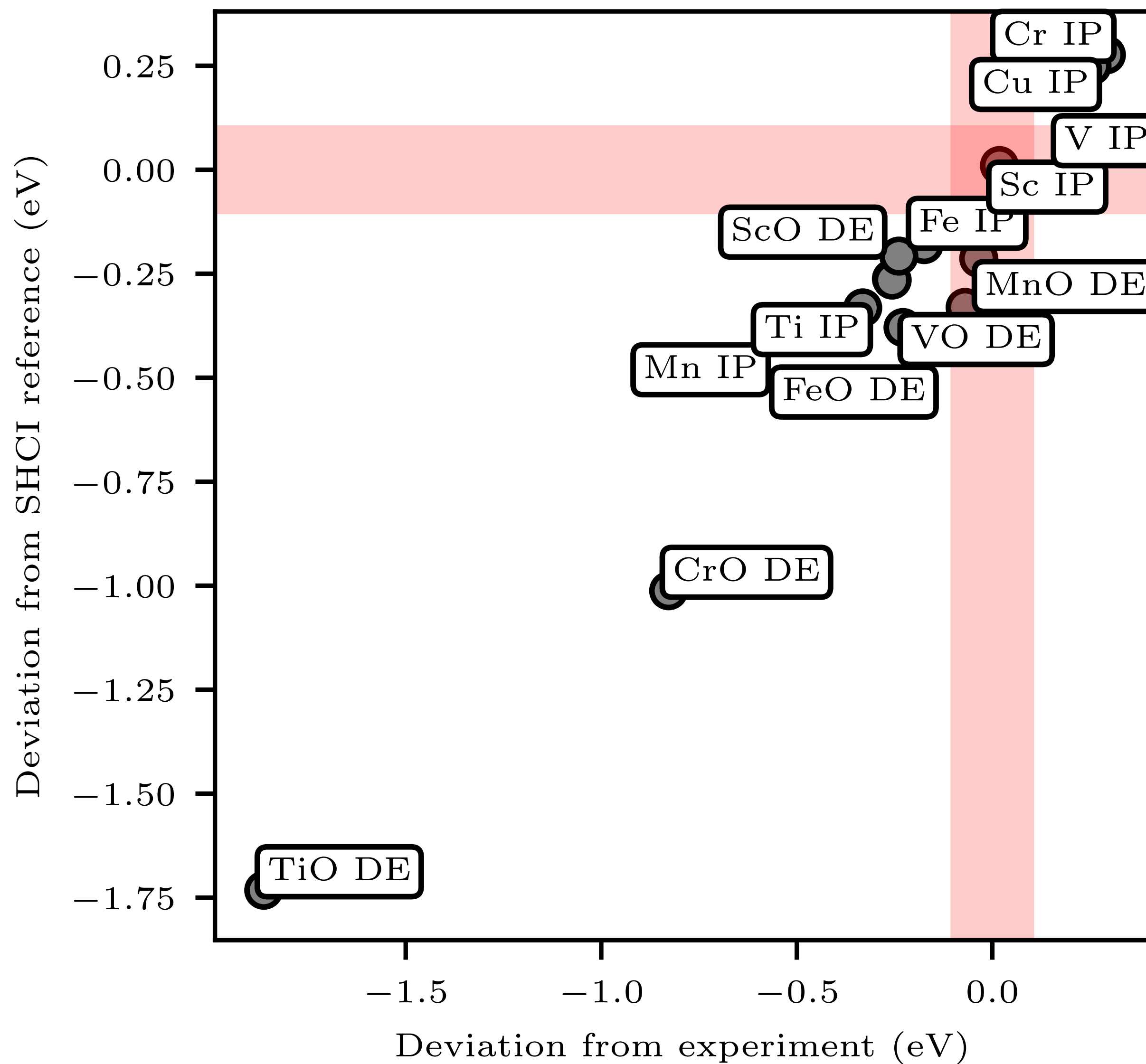


- Provenance well-defined
- Internal reference, precise results on systems containing up to 25 correlated electrons

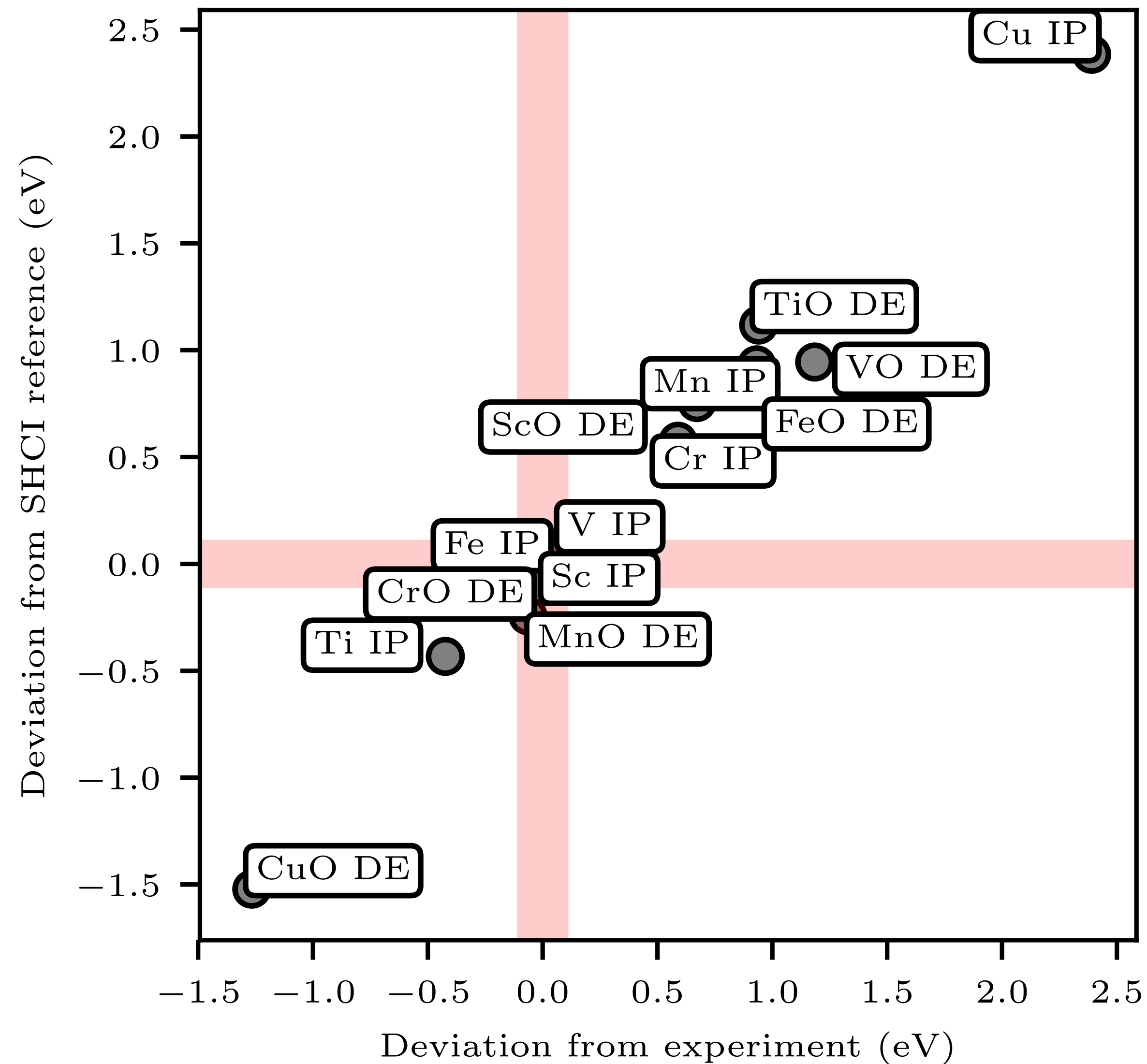
- Enables clean method assessment
- Practical calculations rely on cancellation of error to obtain highly accurate results



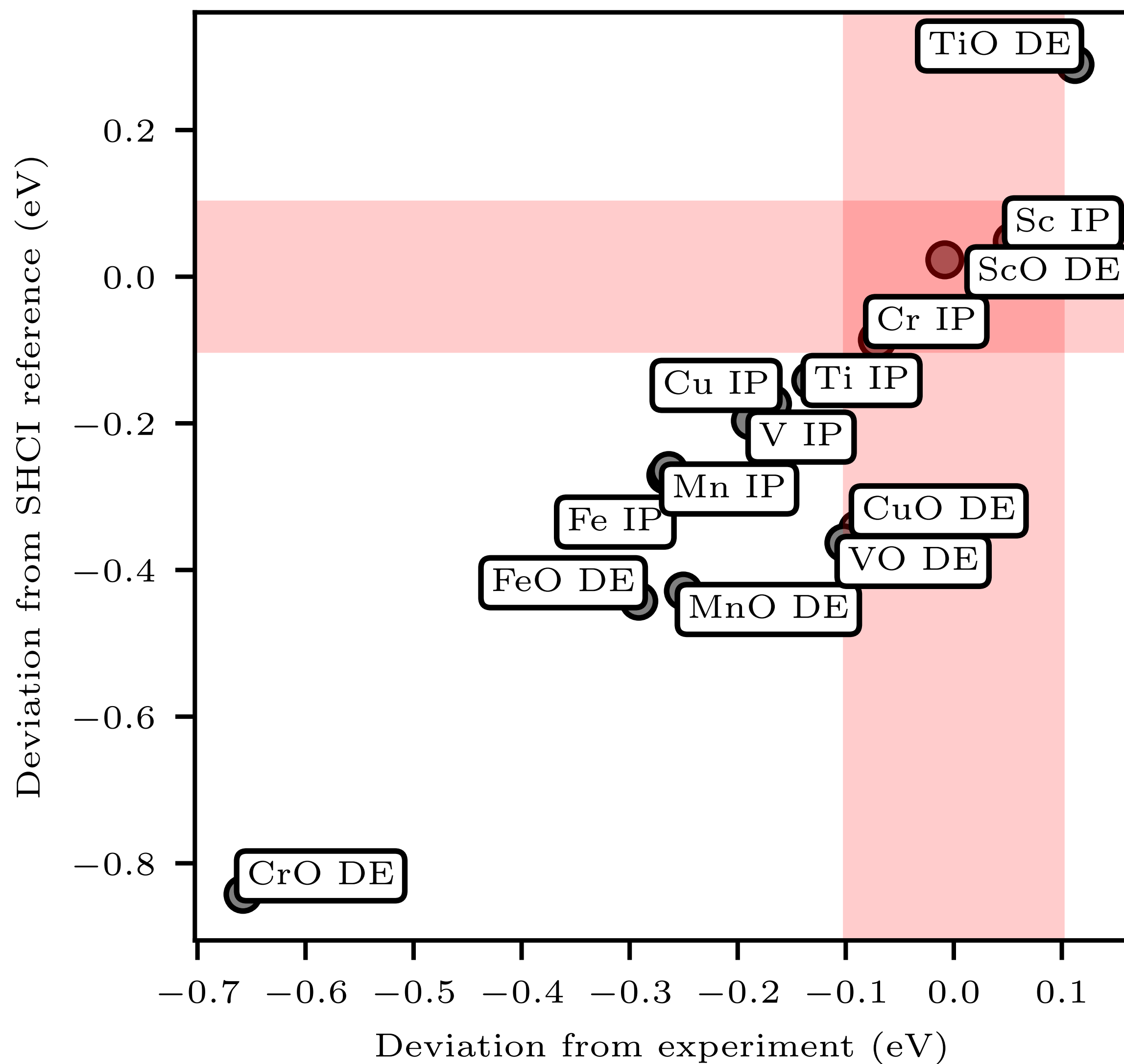
HSE06



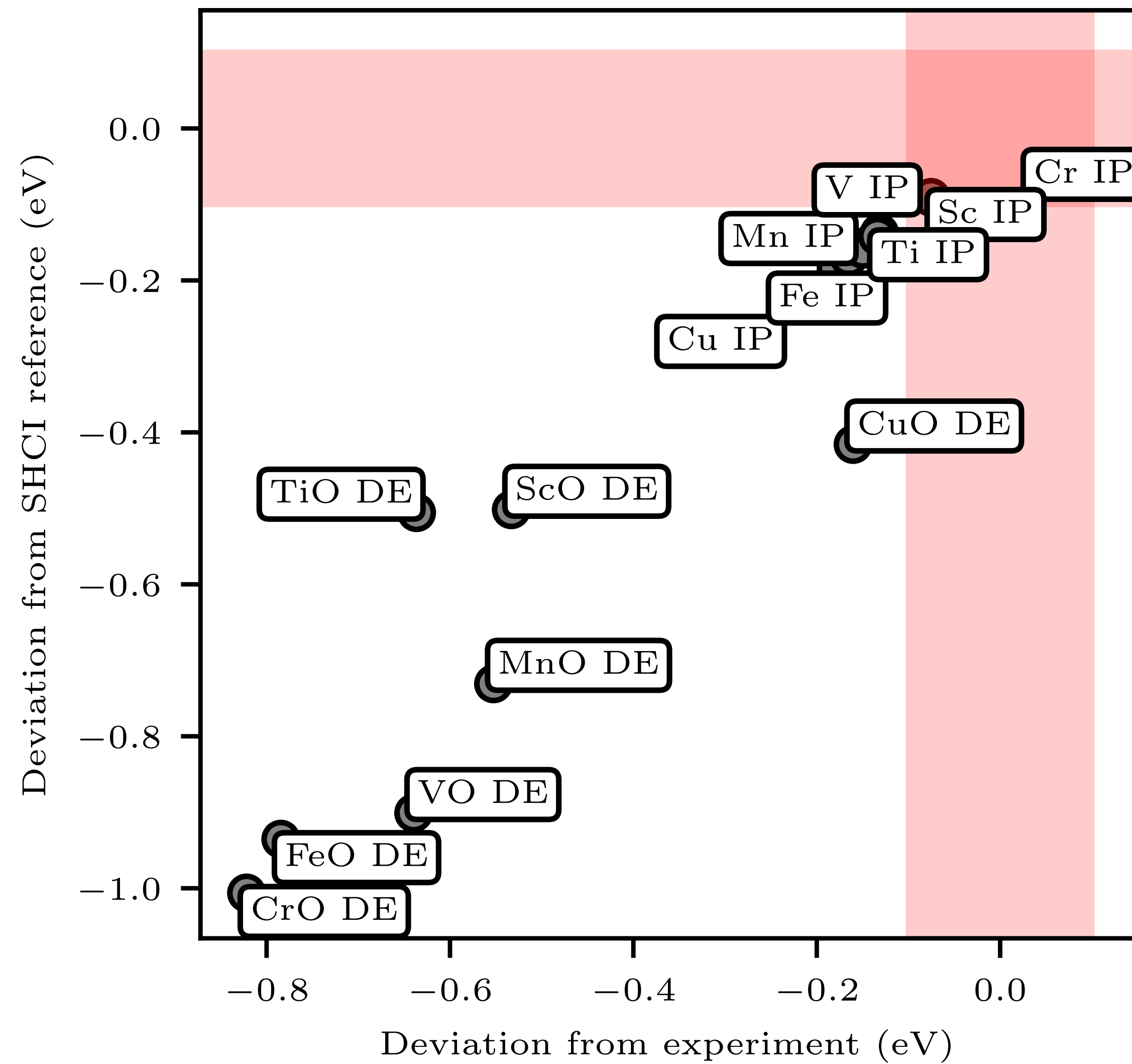
PBE



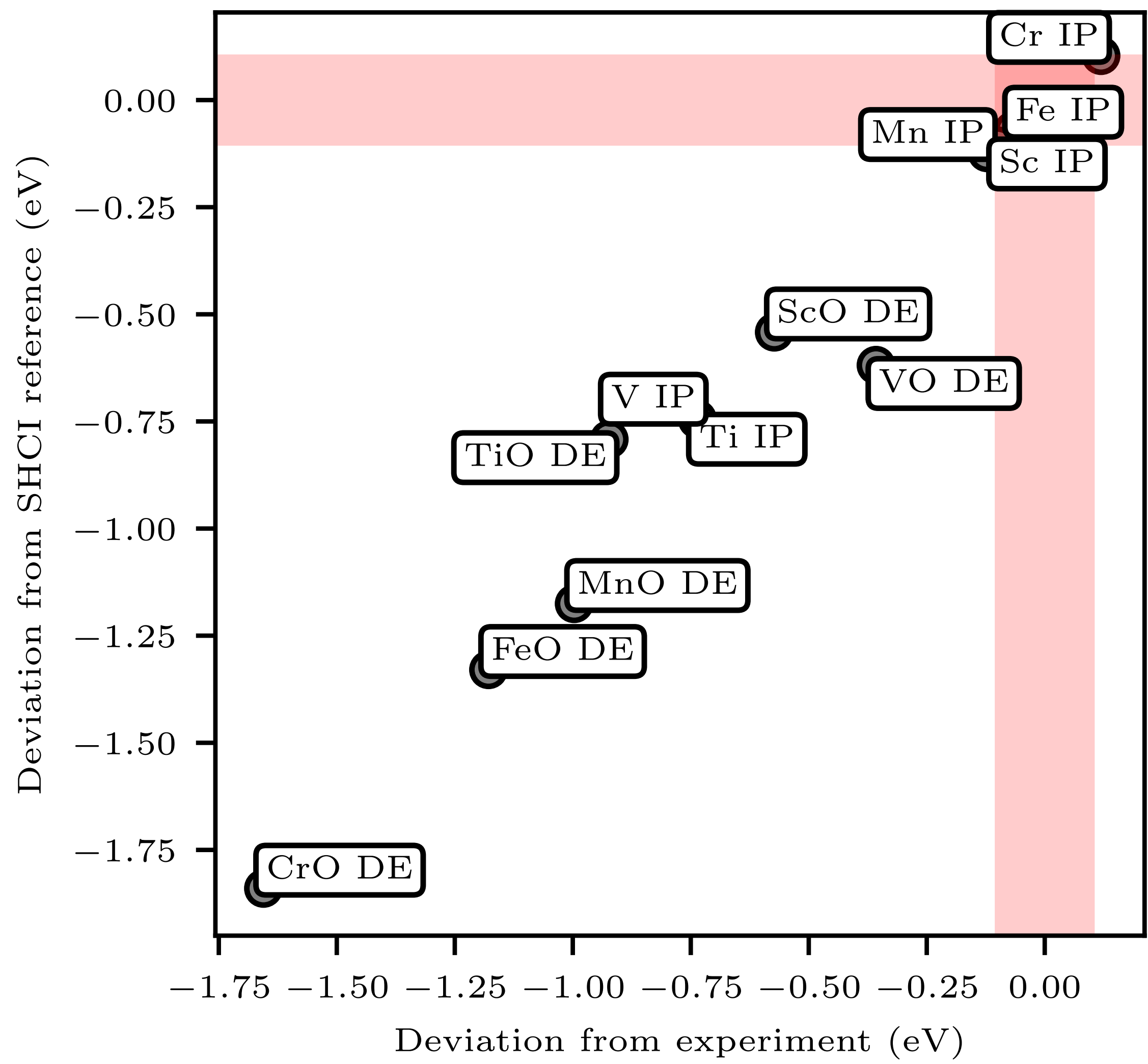
DMC(SD)



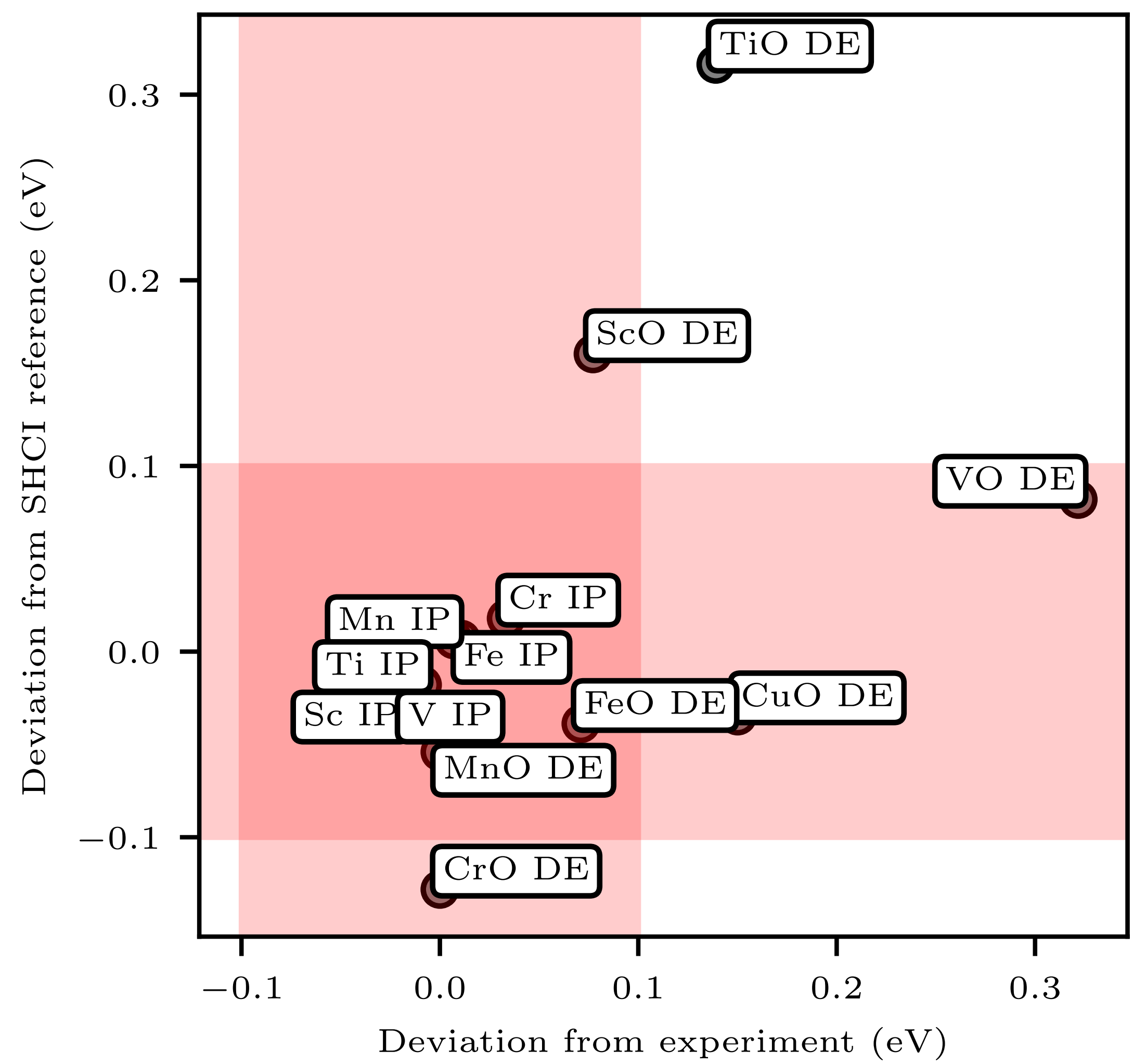
CCSD



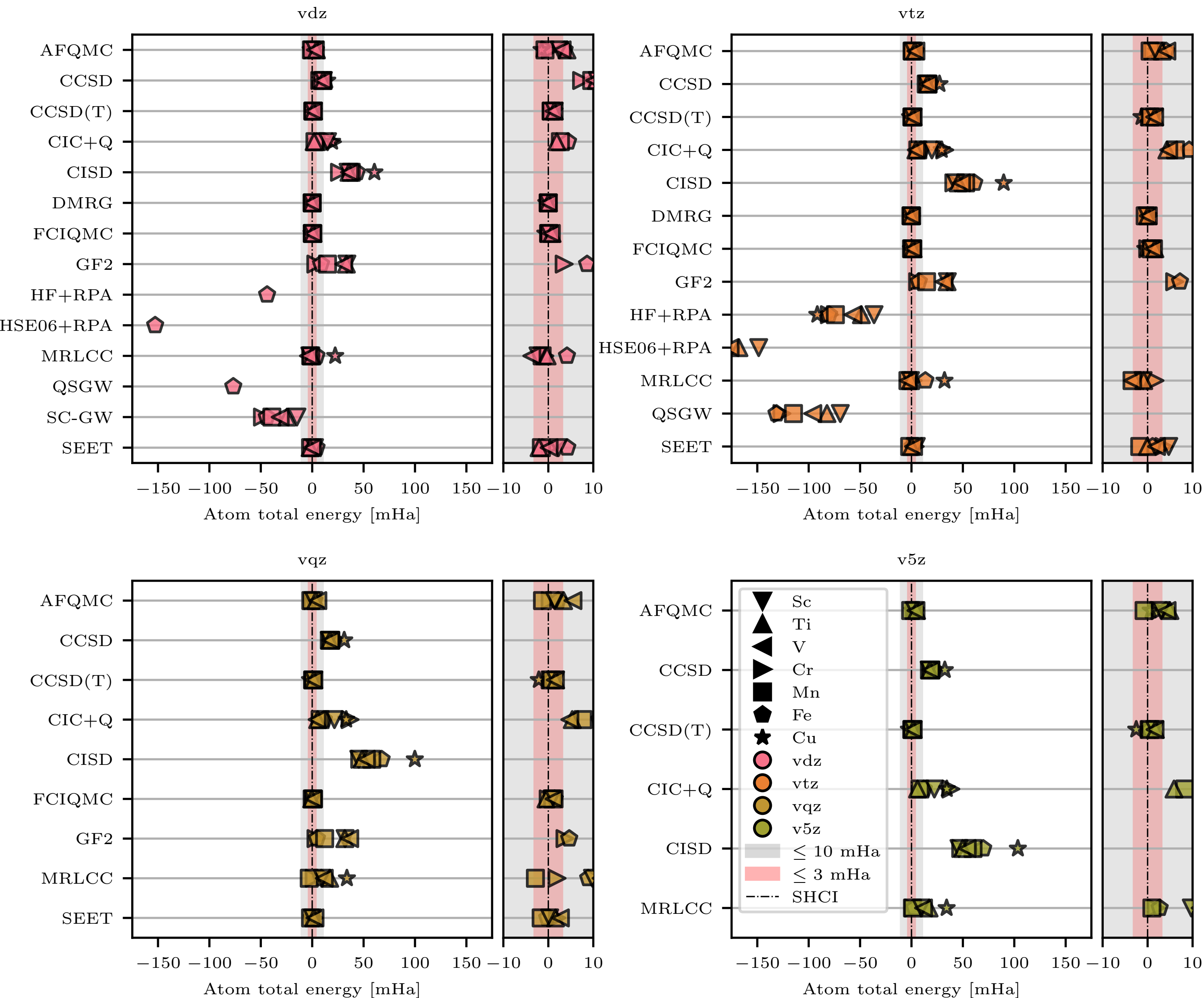
GF2



AFQMC



Total energy in a basis



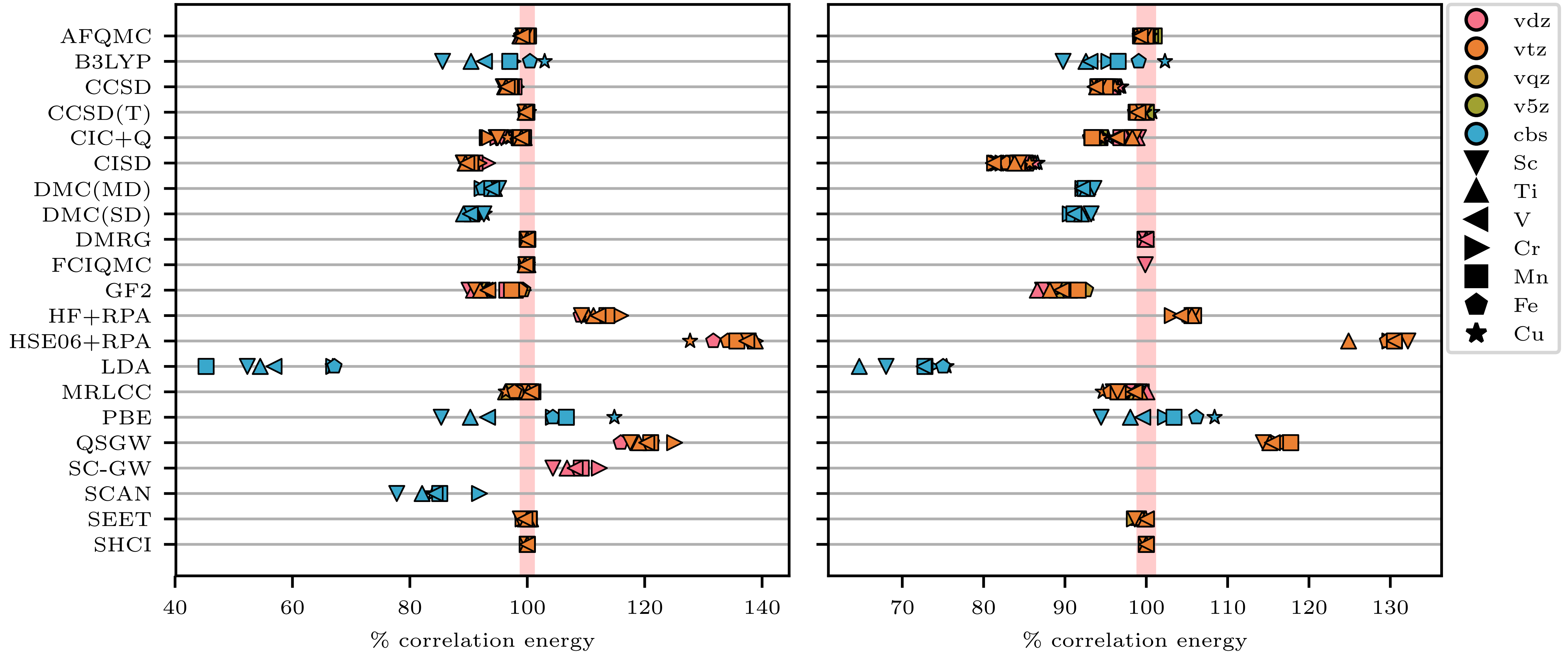
Error is **only** the error in the solution method.

Useful for debugging and improving methods

Correlation energy

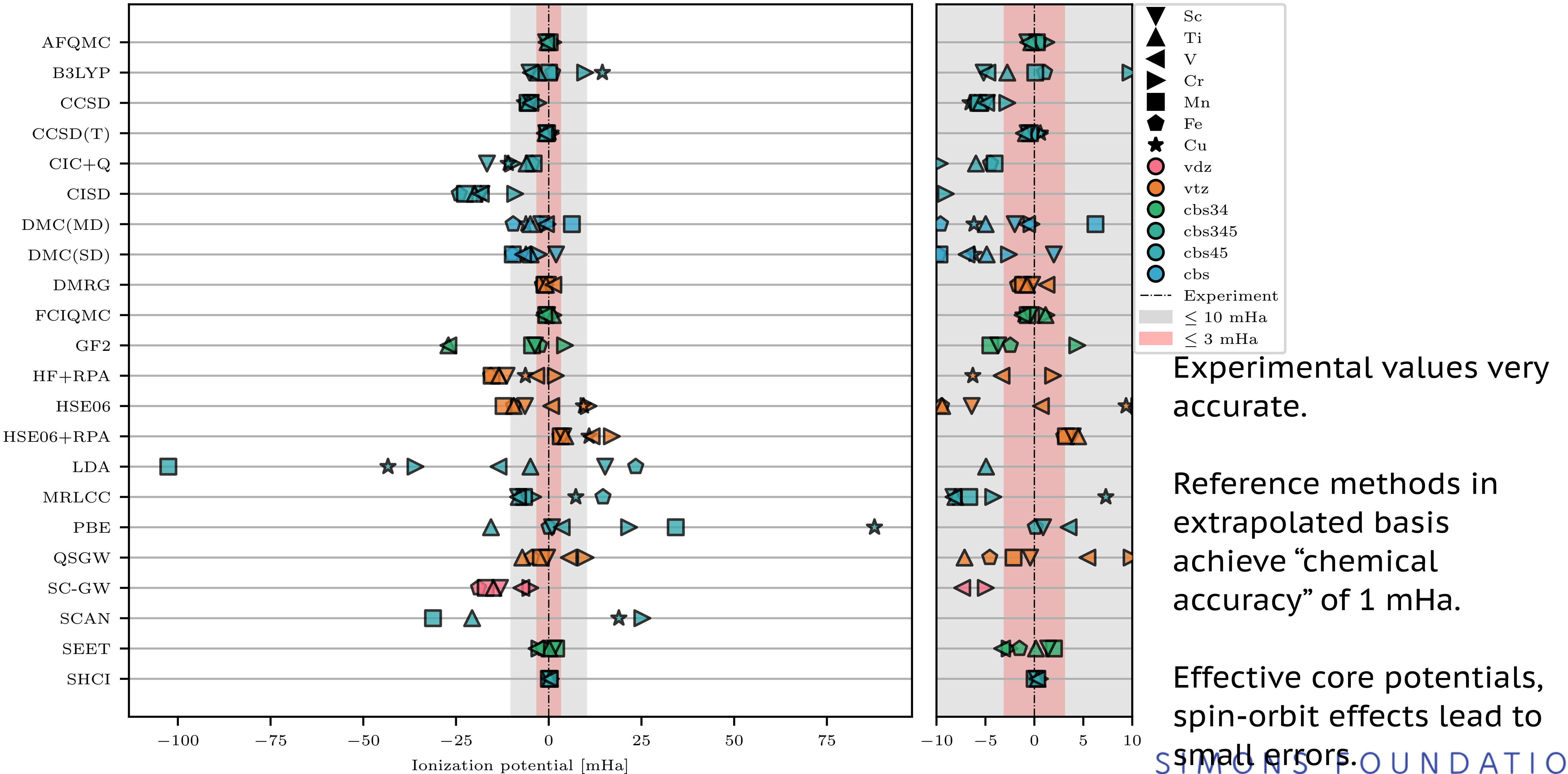
(a) Atoms

(b) TM-O molecules

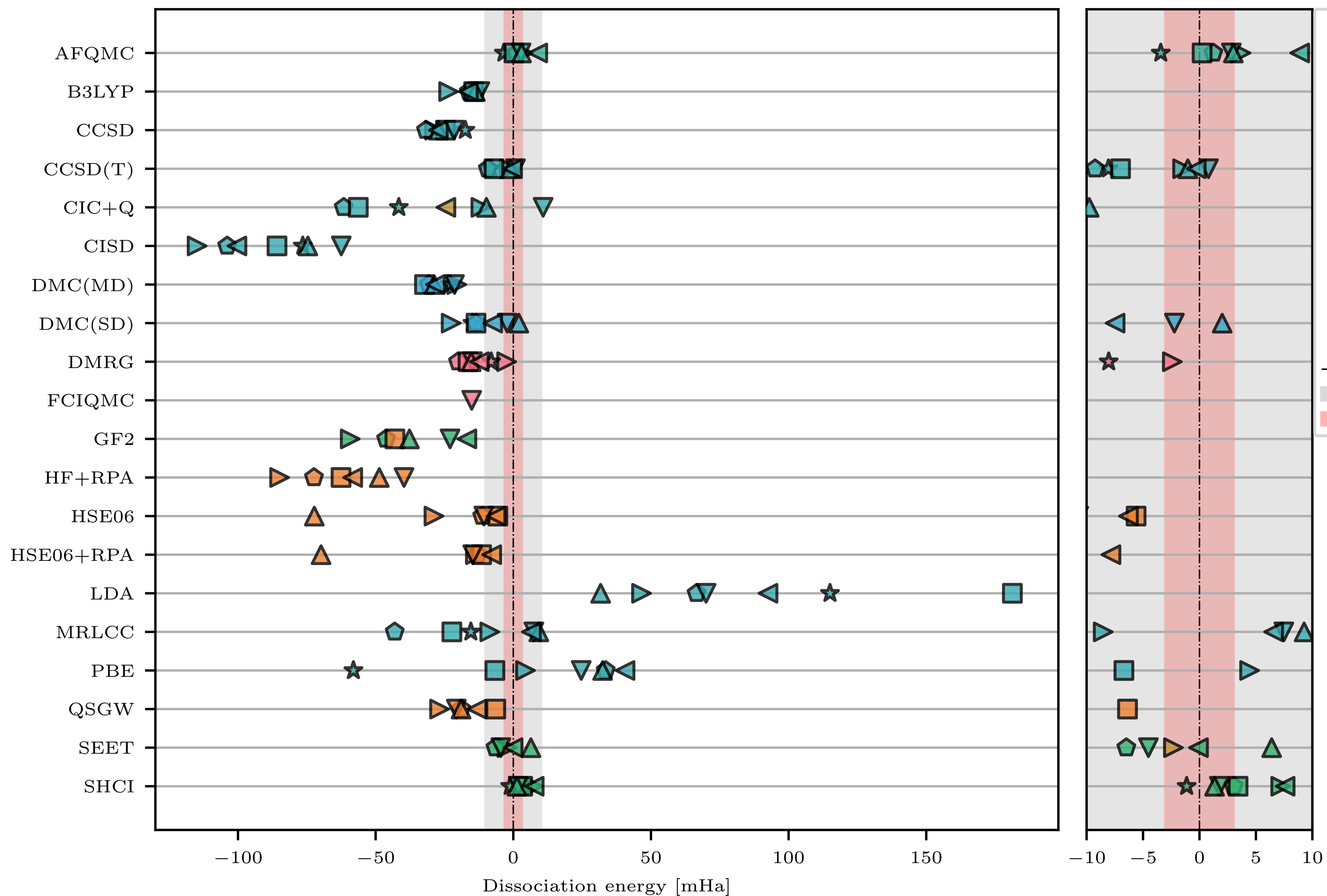


Many methods obtain very similar percentage of correlation energy across different systems and basis sets

Ionization relative to experiment



Disociation energy



Experimental values less precise (10 mHa spread)

Reference methods roughly as good as experiment

Effective core potentials, spin-orbit effects lead to small errors.

Did we test “strong correlation”

The systems we tested were not what chemists would call “strongly correlated.”

The 1-RDM for an accurate result is not too different from the 1-RDM of a single determinant wave function.

However, so-called “dynamic correlation” is very large in these systems.

This is why methods like CCSD(T) work very well but multi-reference methods perform worse; they have “wasted” some of their description space.

Other thoughts

One can implement correction methods based on the consistent performance of some methods (FN-DMC, for example)